FORMULAS WITH POLYTOMIC IONS

Matching the horizontal and vertical axes, write the formulas of the compounds with the following combination of ions.

| | он: | он, | CO,* | SO, | PO ₁ 3 |
|------|---------------------|-----------------------------------|---|---|---|
| H | HOH (H₂O) | HNO ₃ | H₂CO₃ | H ₂ SO ₄ | H ₃ PO ₄ |
| Na' | NaOH | NaNO ₃ | Na₂CO₃ | Na₂SO₄ | Na ₃ PO ₄ |
| Mg*° | Mg(OH)₂ | Mg(NO ₃) ₂ | MgCO ₃ | MgSO ₄ | Mg ₃ (PO ₄) ₂ |
| ZH* | NH₄OH | NH ₄ NO ₃ | (NH ₄) ₂ CO ₃ | (NH ₄) ₂ SO ₄ | (NH ₄) ₃ PO ₄ |
| Ca** | кон | KNO ₃ | K₂CO₃ | K₂SO₄ | K ₃ PO ₄ |
| κ. | Al(OH) ₃ | Al(NO ₃) ₃ | Al ₂ (CO ₃) ₃ | Al ₂ (SO ₄) ₃ | AIPO ₄ |
| Al*3 | РЬ(ОН)4 | Pb(NO ₃) ₄ | Pb(CO ₃) ₂ | Pb(SO ₄) ₂ | Pb ₃ (PO ₄) ₄ |

Polyatomic Ionic Formulas Answers

Patrick Flores-Scott

Polyatomic Ionic Formulas Answers:

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