

- a) What are Carbenes ? What is the state of their hybridisation ? Discuss structure and stability of singlet and triplet dichlorocarbene. 5
- b) Briefly explain the following observations :  $2 \times 2$
- Phenyl anion is more stable than cyclohexyl anion.
  - N, N-dimethylaniline is weak as a base than N, N, Z, 6 - tetramethylaniline.
- c) Methane reacts with  $\text{Cl}_2$  to form  $\text{CH}_3\text{Cl}$  and  $\text{HCl}$ , but with  $\text{I}_2$  doesn't occur at appreciable rate, even  $\text{I} - \text{I}$  bond is much weaker than  $\text{Cl} - \text{Cl}$  bond. Explain why this is SO. 3
- d) What products are formed when glycine ( $\text{H}_2\text{NCH}_2\text{CO}_2\text{H}$ ) is treated with conc.  $\text{HCl}$  and  $\text{NaOH}$  separately. 3
2. a) Draw the structure for each of the following compounds :  $1\frac{1}{2} \times 4$
- (1R, 3S) - 1, 3-cyclohexanediol
  - (1Z, 3E) - 1-bromo-1, 3-pentadiene.

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