

**Chemistry: pH and pOH calculations**

**Part 1:** Fill in the missing information in the table below.

pH	[ H <sub>3</sub> O <sup>+</sup> ]	pOH	[ OH <sup>-</sup> ]	ACID or BASE?
3.78	$1.66 \times 10^{-4} \text{ M}$	10.22	$6.03 \times 10^{-11} \text{ M}$	Acid
3.41	$3.89 \times 10^{-4} \text{ M}$	10.59	$2.57 \times 10^{-11} \text{ M}$	Acid
8.81	$1.55 \times 10^{-9} \text{ M}$	5.19	$6.46 \times 10^{-6} \text{ M}$	Base
8.69	$2.04 \times 10^{-9} \text{ M}$	5.31	$4.88 \times 10^{-6} \text{ M}$	Base
8.46	$3.47 \times 10^{-9} \text{ M}$	5.54	$2.88 \times 10^{-6} \text{ M}$	Base
12.1	$8.45 \times 10^{-13} \text{ M}$	1.90	$1.26 \times 10^{-2} \text{ M}$	Base
11.86	$1.38 \times 10^{-12} \text{ M}$	2.14	$7.24 \times 10^{-3} \text{ M}$	Base
3.40	$3.98 \times 10^{-4} \text{ M}$	10.6	$2.31 \times 10^{-11} \text{ M}$	Acid
10.91	$1.23 \times 10^{-11} \text{ M}$	3.09	$8.13 \times 10^{-5} \text{ M}$	Base
5.13	$7.49 \times 10^{-6} \text{ M}$	8.87	$1.35 \times 10^{-9} \text{ M}$	Acid
4.06	$8.71 \times 10^{-5} \text{ M}$	9.94	$1.15 \times 10^{-10} \text{ M}$	Acid
6.41	$3.89 \times 10^{-7} \text{ M}$	7.59	$2.57 \times 10^{-8} \text{ M}$	Acid
4.16	$6.92 \times 10^{-5} \text{ M}$	9.84	$1.45 \times 10^{-10} \text{ M}$	Acid
0.98	$1.06 \times 10^{-5} \text{ M}$	13.0	$1.00 \times 10^{-13} \text{ M}$	Acid
10.18	$6.61 \times 10^{-11} \text{ M}$	3.82	$1.51 \times 10^{-4} \text{ M}$	Base
7.93	$1.17 \times 10^{-8} \text{ M}$	6.07	$8.53 \times 10^{-7} \text{ M}$	Base
7.05	$8.91 \times 10^{-8} \text{ M}$	6.95	$1.12 \times 10^{-7} \text{ M}$	-Base
9.33	$4.73 \times 10^{-10} \text{ M}$	4.67	$2.14 \times 10^{-5} \text{ M}$	Base
12.67	$2.14 \times 10^{-13} \text{ M}$	1.33	$4.68 \times 10^{-2} \text{ M}$	Base
12.0	$1.0 \times 10^{-12} \text{ M}$	2.01	$9.87 \times 10^{-3} \text{ M}$	Base
11.68	$2.09 \times 10^{-12} \text{ M}$	2.32	$4.79 \times 10^{-3} \text{ M}$	Base
7.04	$9.22 \times 10^{-8} \text{ M}$	6.96	$1.10 \times 10^{-7} \text{ M}$	-Base
1.76	$1.74 \times 10^{-2} \text{ M}$	12.24	$5.75 \times 10^{-13} \text{ M}$	Acid
2.70	$2.00 \times 10^{-3} \text{ M}$	11.3	$5.39 \times 10^{-12} \text{ M}$	Acid

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**Jin-Ying Zhang**



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### **Table of Contents Ph And Poh Chemistry If8766 Answers**

1. Understanding the eBook Ph And Poh Chemistry If8766 Answers
  - The Rise of Digital Reading Ph And Poh Chemistry If8766 Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Ph And Poh Chemistry If8766 Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Ph And Poh Chemistry If8766 Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Ph And Poh Chemistry If8766 Answers

- Personalized Recommendations
  - Ph And Poh Chemistry If8766 Answers User Reviews and Ratings
  - Ph And Poh Chemistry If8766 Answers and Bestseller Lists
5. Accessing Ph And Poh Chemistry If8766 Answers Free and Paid eBooks
    - Ph And Poh Chemistry If8766 Answers Public Domain eBooks
    - Ph And Poh Chemistry If8766 Answers eBook Subscription Services
    - Ph And Poh Chemistry If8766 Answers Budget-Friendly Options
  6. Navigating Ph And Poh Chemistry If8766 Answers eBook Formats
    - ePub, PDF, MOBI, and More
    - Ph And Poh Chemistry If8766 Answers Compatibility with Devices
    - Ph And Poh Chemistry If8766 Answers Enhanced eBook Features
  7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Ph And Poh Chemistry If8766 Answers
    - Highlighting and Note-Taking Ph And Poh Chemistry If8766 Answers
    - Interactive Elements Ph And Poh Chemistry If8766 Answers
  8. Staying Engaged with Ph And Poh Chemistry If8766 Answers
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Ph And Poh Chemistry If8766 Answers
  9. Balancing eBooks and Physical Books Ph And Poh Chemistry If8766 Answers
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Ph And Poh Chemistry If8766 Answers
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine Ph And Poh Chemistry If8766 Answers
    - Setting Reading Goals Ph And Poh Chemistry If8766 Answers
    - Carving Out Dedicated Reading Time
  12. Sourcing Reliable Information of Ph And Poh Chemistry If8766 Answers

- Fact-Checking eBook Content of Ph And Poh Chemistry If8766 Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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