

Phase Diagram POGIL

Name _____ Date _____

Model 1

PHASE DIAGRAM



A phase diagram is a graphical tool to summarize the conditions under which equilibrium exists between the different states of matter. It also allows us to predict the phase of a substance that is, under a set of given temperature and pressure.

Critical Thinking Questions

1. What label is on the y-axis?
2. What label is on the x-axis?
3. Label the three phases of matter that are in the diagram.
4. Are there points that all three phases can be in equilibrium?
5. The point where the solid, liquid, and gas lines meet, the triple point is _____.
6. The temperature where the liquid-gas line meets the 1 atm pressure line, the normal boiling point is _____.
7. The point where the solid-liquid line meets the 1 atm pressure line, the normal melting point is _____.

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