

# States of Matter and Phase Changes

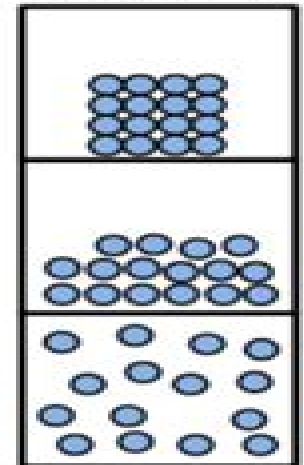
Matter can exist in three major states:

➔ **Solid** → Low temperature and low energy.  
Has a definite volume and a definite shape.

➔ **Liquid** → Takes the shape of its container.  
Has a definite volume, but no definite composition.

➔ **Gas** → High amount of energy and disorder.  
Has no definite shape and no definite composition.

Draw each state below:



Instructions: Identify the phase changes below and whether energy is increasing or decreasing.

1)		Energy ➔ Increase <del>or</del> decrease		Phase Change: <b>Freezing</b>
2)		Energy ➔ Increase <del>or</del> decrease		Phase Change: <b>Condensation</b>
3)		Energy ➔ Increase <del>or</del> decrease		Phase Change: <b>Melting</b>

# Phases Of Matter Guided Answers

**Wendy Conklin**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

## **Phases Of Matter Guided Answers:**

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**Physical and Chemical Changes** Edward P. Ortleb,Richard Cadice,1993-09-01 General chemistry information including everything from matter to radioactivity For grades 5 to 9

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