

ACTIVITY: Molecular Polarity Computer Simulation (PhET)

Go to: <https://phet.illinois.edu/> Click on the arrow, DOWNLOAD, OPEN and RUN the MOLECULE POLARITY program. Or google "phet molecule polarity" to find the simulation.

Introduction:

In this atomic level simulation, you will investigate how atoms' **electronegativity** values affect the bonds they produce. When two atoms form a covalent bond, a pair of electrons is shared between atoms. **Electronegativity** is a measure of a single atom's ability to attract electrons shared in that bond. In this lab you will work at your own pace, to answer a number of questions. To begin, from what you've already learned about the protons and electrons in an atom, what would cause an atom to have a **high** electronegativity value?

Why might an atom have a low electronegativity value?

Procedure:

New Molecule

- ☛ Turn on (check) all view options. **distort** our gaze the molecule with your mouse.
- ☛ Take your time and investigate how the binary compound's bond behaves when the atom's electronegativity and orientation are changed. **Do not rush through this step.**

Describe the bond formed between two atoms with **similar, low** electronegativities.

Describe the bond formed between two atoms with **similar, high** electronegativities.

Describe the bond formed between two atoms with **very different** electronegativities.

Describe (in your own words) what is meant by **partial charges**, δ^- and δ^+ .

δ^- represents: _____

δ^+ represents: _____

What happens when the electric field is applied to a very polar molecule?

Why do you think this is? _____

View

- ☒ Bond Dipoles **ON**
- ☒ Partial Charges
- ☒ Bond Orientation

Surfaces

- ☐ None
- ☐ Electrostatic Potential
- ☒ Electron Density

Electric Field

- ☒ ON ☐ OFF



Phet Molecular Polarity Lab Answer Key

Nivaldo J. Tro



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Phet Molecular Polarity Lab Answer Key Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has become much more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Phet Molecular Polarity Lab Answer Key**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we will delve to the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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