

Exploring nuclear decay – Beta Decay

- Go to PhET Simulations. Find and open the sim called Beta Decay. http://phet.colorado.edu/simulations/sims.php?sim=Beta_Decay
- Click on "Run Now" to start the simulation. When the simulation opens click on the "Single Atom" Tab.

After the simulation begins, follow the instructions below and answer the questions that follow each instruction.

1. Watch the Hydrogen-3 (${}^3\text{H}$) nucleus until it decays. Click 'Reset Nucleus' and watch it again. Repeat this at least 8 times. List at least 3 observations that compare the ${}^3\text{H}$ nucleus before and after it decays.
2. Click on the 'Reset Nucleus' button and run the program again, but this time click the 'Pause' button  immediately after a decay so you can look at what comes flying out of the nucleus. Describe what you observe.
3. Why does the Hydrogen (${}^3\text{H}$) nucleus become a Helium (${}^3\text{He}$) nucleus after it decays? (You may have to look at a Periodic Table to determine this.)
4. Why doesn't the mass number change after Hydrogen-3 (${}^3\text{H}$) decays to Helium-3 (${}^3\text{He}$)?
5. Select the Carbon-14 (${}^{14}\text{C}$) nucleus and run the simulation. Do this at least 8 times. List at least three observations that compare the Carbon-14 nucleus before and after it decays.
6. Describe the similarities and differences you observed between ${}^3\text{H}$ decay and ${}^{14}\text{C}$ decay.
7. Look at the Decay Time Chart across the top of the screen. It should display the decayed nuclei of all of the previous trials you watched. Describe any pattern for the decay times that are represented on the chart. Does this indicate that the decay of a specific nucleus is a predictable pattern or random?



Phet Simulation Answers For Beta Decay

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Table of Contents Phet Simulation Answers For Beta Decay

1. Understanding the eBook Phet Simulation Answers For Beta Decay
 - The Rise of Digital Reading Phet Simulation Answers For Beta Decay
 - Advantages of eBooks Over Traditional Books
2. Identifying Phet Simulation Answers For Beta Decay
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Phet Simulation Answers For Beta Decay
 - User-Friendly Interface
4. Exploring eBook Recommendations from Phet Simulation Answers For Beta Decay
 - Personalized Recommendations
 - Phet Simulation Answers For Beta Decay User Reviews and Ratings
 - Phet Simulation Answers For Beta Decay and Bestseller Lists
5. Accessing Phet Simulation Answers For Beta Decay Free and Paid eBooks

- Phet Simulation Answers For Beta Decay Public Domain eBooks
 - Phet Simulation Answers For Beta Decay eBook Subscription Services
 - Phet Simulation Answers For Beta Decay Budget-Friendly Options
6. Navigating Phet Simulation Answers For Beta Decay eBook Formats
- ePUB, PDF, MOBI, and More
 - Phet Simulation Answers For Beta Decay Compatibility with Devices
 - Phet Simulation Answers For Beta Decay Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Phet Simulation Answers For Beta Decay
 - Highlighting and Note-Taking Phet Simulation Answers For Beta Decay
 - Interactive Elements Phet Simulation Answers For Beta Decay
8. Staying Engaged with Phet Simulation Answers For Beta Decay
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Phet Simulation Answers For Beta Decay
9. Balancing eBooks and Physical Books Phet Simulation Answers For Beta Decay
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Phet Simulation Answers For Beta Decay
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Phet Simulation Answers For Beta Decay
- Setting Reading Goals Phet Simulation Answers For Beta Decay
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Phet Simulation Answers For Beta Decay
- Fact-Checking eBook Content of Phet Simulation Answers For Beta Decay
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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