(note: "sink" means stays on the bottom) @

Start: 1. Google: "Phet Density sim" and click on the first link OR go to my blog OR go to http://phet.colorado.edu/sims/density-and-buoyancy/density_en.html

- 2. Click on the button
- 3. Experiment with choosing a material:

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	d to	Block .	Albertalia		
0.40 h	grt.				
					-
Marie Street	the material	and the second		349	-

material	S=sink F=Float	Density given		
Styrofoam	F	0.15 kg/L		
Wood	F	0.4 kg/L kg/L		
Loe	F	0.92 kg/L		
Brick	S	2 kg/L		
Aluminum	5	2.7 kg/L		

with your partner about this possibility- can you <u>change the mass</u> of the aluminum block <u>without</u> <u>changing the volume</u> of the aluminum block?

/ Circle one YES/ NO

Try to get aluminum to float.

5. Look at the density triangle at the bottom of the box- Why do you think this does or does not move?



Mass and volume change together (proportionally) so density stays the same for aluminum

6. How does the density of aluminum (2.70 kg/L) help explain what you see?

Tolk

The aluminum will sink in the water because the density of the aluminum is 2.7kg/L and the density of water is 1 kg/L. We have learned that if something is more dense than water, it will sink

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Kathleen Armour

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- 1. Understanding the eBook Phet Density Activity Answer Key
 - The Rise of Digital Reading Phet Density Activity Answer Key
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Phet Density Activity Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Phet Density Activity Answer Key
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Phet Density Activity Answer Key
 - Personalized Recommendations
 - Phet Density Activity Answer Key User Reviews and Ratings
 - Phet Density Activity Answer Key and Bestseller Lists
- 5. Accessing Phet Density Activity Answer Key Free and Paid eBooks

- Phet Density Activity Answer Key Public Domain eBooks
- Phet Density Activity Answer Key eBook Subscription Services
- Phet Density Activity Answer Key Budget-Friendly Options
- 6. Navigating Phet Density Activity Answer Key eBook Formats
 - o ePub, PDF, MOBI, and More
 - Phet Density Activity Answer Key Compatibility with Devices
 - Phet Density Activity Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - o Adjustable Fonts and Text Sizes of Phet Density Activity Answer Key
 - Highlighting and Note-Taking Phet Density Activity Answer Key
 - Interactive Elements Phet Density Activity Answer Key
- 8. Staying Engaged with Phet Density Activity Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Phet Density Activity Answer Key
- 9. Balancing eBooks and Physical Books Phet Density Activity Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Phet Density Activity Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Phet Density Activity Answer Key
 - $\circ\,$ Setting Reading Goals Phet Density Activity Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Phet Density Activity Answer Key
 - Fact-Checking eBook Content of Phet Density Activity Answer Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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