Acids & Bases: Conj. Acid-Base Pairs & "pet block." Calculations Skill Review

Skill - The student will be able to identify the acid, have, conjugate-acid, and conjugate-base in an acid - have reaction.

Label the acid and base on the reactant side of each of the following equations and label the products as the conjugate acid and conjugate base.

1 OH + HOM + H₂O + CM

2 HCOV + OH + HUO + COV+

Bill to the following table

	HWO.	H,O	Conjugate And	Conjugate Save	HERO" + H'O - HO", + H'O.

Write an equation that shows the reaction of ammonia (MHJ) with hydrobronic acid (HBr).
Label the acid, base, conjugate acid & conjugate base.

Skill - Student will be able to calculate with the [H+] or [OH+] from the ion product (Km) or pit from pOH (one-step pH block calculation)

- Calculate the [Onl] of a solution with a [N1] of 1.00x20 ****
- Calculate the (H*) calculation of a solution with a proof 8.50.
- 8. Calculate the pOH of a solution with a pH of 4.63.
- 5: Calculate the pOH of a solution with a 50H Lof 5.62x10 NH.

Marrie

Ph Calculations Answer Key

Andrew Hunt, Graham Curtis, Graham Hill

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