

The letters "pH" represent the French words "pouvoir hydrogene" which means "hydrogen power".

- The definition of pH is *pH is equal to the negative log (logarithm) of the hydrogen ion concentration of a solution.*
- The logarithm of a number is the power to which 10 must be raised to equal that number.

A pH value of less than 7 indicates a(n) ACIDIC solution. A pH value of 7 indicates a neutral solution. A pH value of more than 7 indicates a(n) BASIC solution.

PROBLEMS: Show all work and circle the final answer.

1. Determine the pH of a 0.010 M HNO_3 solution.

$$\begin{aligned} \text{pH} &= -\log[0.010] \\ &= \boxed{2} \end{aligned}$$

2. What is the pH of a 2.5×10^{-6} M solution of HCl?

$$\begin{aligned} \text{pH} &= -\log[2.5 \times 10^{-6}] \\ &= \boxed{5.6} \end{aligned}$$

3. Calculate the pH of a solution of 0.0025M H_2SO_4 .

$$\begin{aligned} \text{pH} &= -\log[2 \times 0.0025] \\ &= -\log[0.005] \\ &= \boxed{2.3} \end{aligned}$$

4. Calculate the pH of a 0.0010 M NaOH solution.

$$\begin{aligned} [\text{H}^+][0.0010] &= 1.0 \times 10^{-14} \\ [\text{H}^+] &= \frac{1.0 \times 10^{-14}}{0.0010} \\ \text{pH} &= -\log[1 \times 10^{-11}] \\ &= \boxed{11} \end{aligned}$$

5. What is the pH of a 0.020M $\text{Sr}(\text{OH})_2$ solution?

$$\begin{aligned} [\text{OH}^-][0.04] &= 1 \times 10^{-14} \\ [\text{OH}^-] &= \frac{1 \times 10^{-14}}{0.04} \\ \text{pH} &= -\log[\text{H}^+] \\ &= -\log 2.5 \times 10^{-9} \end{aligned}$$

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Suzanne Nielsen,2014-09-04 This book provides information on the techniques needed to analyze foods in laboratory experiments All topics covered include information on the basic principles procedures advantages limitations and applications This book is ideal for undergraduate courses in food analysis and is also an invaluable reference to professionals in the food industry General information is provided on regulations standards labeling sampling and data handling as background for chapters on specific methods to determine the chemical composition and characteristics of foods Large expanded sections on spectroscopy and chromatography also are included Other methods and instrumentation such as thermal analysis ion selective electrodes enzymes and immunoassays are covered from the perspective of their use in the analysis of foods A website with related teaching materials is accessible to instructors who adopt the textbook

AQA A2 Chemistry Student Unit Guide: Unit 4

Kinetics, Equilibria and Organic Chemistry Margaret Cross,2010-02-26 Student Unit Guides are perfect for revision Each guide is written by an examiner and explains the unit requirements summarises the relevant unit content and includes a series of specimen questions and answers There are three sections to each guide Introduction includes advice on how to use the guide an explanation of the skills being tested by the assessment objectives an outline of the unit or module and

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Chemistry Trace Jordan, Neville R. Kallenbach, 2017 Chemistry The Molecules of Life offers chemical insights within the context of health pharmaceuticals and the function of biological molecules The contextualized presentation of topics gives students a broad introduction to chemistry and helps them to see the relevance of chemistry to their personal lives

Environmental Organic Chemistry René P. Schwarzenbach, Philip M. Gschwend, Dieter M. Imboden, 2005-06-24 Environmental Organic Chemistry focuses on environmental factors that govern the processes that determine the fate of organic chemicals in natural and engineered systems The information discovered is then applied to quantitatively assessing the environmental behaviour of organic chemicals Now in its 2nd edition this book takes a more holistic view on physical chemical properties of organic compounds It includes new topics that address aspects of gas solid partitioning bioaccumulation and transformations in the atmosphere Structures chapters into basic and sophisticated sections Contains illustrative examples problems and case studies Examines the fundamental aspects of organic physical and inorganic chemistry applied to environmentally relevant problems Addresses problems and case studies in one volume

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Fundamentals of Chemistry Laboratory Studies Frank Brescia, 2012-12-02 Fundamentals of Chemistry Laboratory Studies focuses on the techniques involved in chemical laboratory operations Divided into 13 parts the manual gives

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