

PHET LAB: Ionic Bonds

Name: _____

PHETLABS suggest that some students preconstruct concepts such as ionic bonds, but you should not over-estimate this. For background on PHET's "background" information, see <https://phet.colorado.edu/en/help-center/section-4/phet-lab-101>.

Open the simulation at <https://phet.colorado.edu/en/simulation/ionic-bonds>.

Take 10 minutes to freely explore the tool.

Investigating Ionic Bonding Activity

1. Place sodium chloride (NaCl) in the simulation. Observe what is shown in the simulation. Do other ionic bonds provide NaCl similar behavior to what you are observing? Write a description of these similarities.

2. Experiment with particles against the transfer. Do electrons flow during these ionic bonds?

3. Experiment in simulation tool using the (transfer) button.



Complete the following table using the simulation, using <https://phet.colorado.edu/en/simulation/ionic-bonds>.

4. Hydrogen (H): $1s^1$ 5. Lithium (Li): $1s^2 2s^1$

6. Calcium (Ca): $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2$

Investigating Ions in the Real World Activity

1. Consider the following table which lists some of the essential elements which are part of all life on Earth. Do you see any other elements represented there? Write below with images.



Phet Colorado Lab Answers Beta Decay

H. Behrens, J. Jänecke



Phet Colorado Lab Answers Beta Decay:

Beta Decay Sigmund Flügge, 1955 **Numerical Tables for Beta-decay and Electron Capture** H. Behrens, J. Jänecke, 1969

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