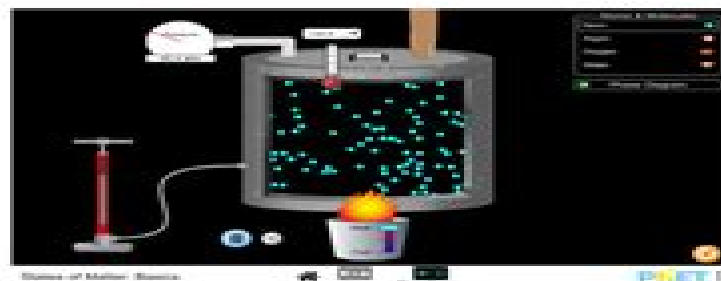


Gas Laws Simulation Lab

- Go to Phet.colorado.edu
- Search for the "States of Matter: Basics" simulation.
- Click on the "Phase Changes" icon.
- Select "Oxygen."



1. The simulation shows a quantity of oxygen molecules. What state of matter is this?
2. How do you know?
Particles have a define shape
3. Heat the oxygen to about 150 K. What state of matter is this?
4. How do you know?
because they are all spread out and have no define shape

Experiment One: Constant volume.

- Leave the oxygen at 150 K. In the table below, record the pressure and temperature of the oxygen in the first row.
- Heat the oxygen to about 450 K.
- Record the new temperature and pressure in the second row of the table.
- Heat the oxygen to about 900 K.
- Record the new temperature and pressure in the third row of the table.

Temperature vs. pressure at constant volume	
Temperature (K)	Pressure (atm)
150	2.5
450	11.8
900	35.5

5. Write a statement describing how the temperature of a quantity of a gas is related to its pressure at constant volume.
As the temperture rises, so does pressure

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O García

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name O García.

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Table of Contents Phet Gas Law Simulation Lab Answer Key

1. Understanding the eBook Phet Gas Law Simulation Lab Answer Key
 - The Rise of Digital Reading Phet Gas Law Simulation Lab Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Phet Gas Law Simulation Lab Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Phet Gas Law Simulation Lab Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Phet Gas Law Simulation Lab Answer Key

- Personalized Recommendations
- Phet Gas Law Simulation Lab Answer Key User Reviews and Ratings
- Phet Gas Law Simulation Lab Answer Key and Bestseller Lists
- 5. Accessing Phet Gas Law Simulation Lab Answer Key Free and Paid eBooks
 - Phet Gas Law Simulation Lab Answer Key Public Domain eBooks
 - Phet Gas Law Simulation Lab Answer Key eBook Subscription Services
 - Phet Gas Law Simulation Lab Answer Key Budget-Friendly Options
- 6. Navigating Phet Gas Law Simulation Lab Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Phet Gas Law Simulation Lab Answer Key Compatibility with Devices
 - Phet Gas Law Simulation Lab Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Phet Gas Law Simulation Lab Answer Key
 - Highlighting and Note-Taking Phet Gas Law Simulation Lab Answer Key
 - Interactive Elements Phet Gas Law Simulation Lab Answer Key
- 8. Staying Engaged with Phet Gas Law Simulation Lab Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Phet Gas Law Simulation Lab Answer Key
- 9. Balancing eBooks and Physical Books Phet Gas Law Simulation Lab Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Phet Gas Law Simulation Lab Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Phet Gas Law Simulation Lab Answer Key
 - Setting Reading Goals Phet Gas Law Simulation Lab Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Phet Gas Law Simulation Lab Answer Key

- Fact-Checking eBook Content of Phet Gas Law Simulation Lab Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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