Lab POGIL: Metal Reactivity - Introduction & Pre-Lab

Discovery Question: How can trends in metal reactivity be used to predict the products of chemical reactions?

Objectives:

- Develop a procedure to test the reactivity of metals.
- List metals in order of reactivity based on the results of your procedure
- Predict the products of chemical reactions by considering the relative reactivity of the reactants

You have been provided with vials containing 3 types of metals (A. B. C). You also have vials tilled with solutions containing cations of each metal (A", B", C"), and an additional vial containing a solution of cation D". You will also have (6-well x 4-well) plates available for small-scale reactions of metals and ionic solutions.

Elemental	Solutions	
Metals	Containing Metal Cations	
A B C	A (contains A**) B (contains B**) C (contains C**) D (contains D**)	

Remember to wear goggles at all times. Avoid skin contact with all solutions. If contact occurs, rinse with water immediately. Solution D will stain skin and clothes.

Pre-lab Requirements:

 Describe a method for predicting charges of ions formed from single atoms of an element using the periodic table.

Use a periodic table to predict charges for ions formed from each element listed below:

Element lon	Element	Lon	Element	Lon
Lithium (Li)	Beryllium (Be)		Copper (Cu)	_/
Sodium (Na)	Magnesium (Mg)	-	Zimc (Zn)	
Potassium (K)	Calcium (Ca)		Silver (Ag)	
Nitrogen (N)	Oxygen (O)	-	Fluorine (F)	
Phosphorous (P)	Sulfur (S)		Chlorine (CI)	
	Selenium (Se)		Bromine (Br)	

Reactivity Of Metals Pogil Answers

RC Schank

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