

Section 1 Review page 118

1. **Describe Democritus's atomic theory.**
Everything in the universe is made up of atoms.
2. **Summarize the main ideas of Dalton's theory.**
 - a. Elements are made up of tiny unique particles called atoms.
 - b. Atoms cannot be divided.
 - c. Atoms of the same element are identical.
 - d. Atoms of different elements can join to form molecules.
3. **Explain why Dalton's theory was more successful than Democritus's theory.**
Dalton had experimental evidence to support his theory, while Democritus did not.
4. **Compare Thomson's atomic model with Rutherford's atomic model.**
Thomson's model- electrons were embedded in a sphere of positive charge
Rutherford's model- positive charge is concentrated in the nucleus and electrons surrounded the nucleus.
5. **Analyzing elements**
 - a. **How did the cathode-ray tube experiment lead to the conclusion that atoms contain electrons?** The air was removed from the tube and the beam of negative particles was observed. This suggests that atoms are made up of smaller units with negative charges.
 - b. **How did the gold-foil experiment lead to the conclusion that the atom has a nucleus?** the positive particles were deflected at large angles and therefore this suggests that a positive charge is concentrated in the center of the atom. (nucleus)
6. **Making Inferences: Does the term *indivisible* still describe the atom? Explain.**
No, the atom is not indivisible. It is made up of smaller parts such as the electron.

Section 2 review page 127

1. **List the charge, mass, and location of each of the three subatomic particles found in atoms.** The proton is positive, the neutron is neutral and electrons are negative. Protons and neutrons are found in the nucleus of an atom. The mass of a proton 1850 times greater than that of an electron.
2. **Explain how you can use an atom's mass number and atomic number to determine the number of protons, electrons, and neutrons to determine the number of protons, electrons, and neutrons in the atom.** The atomic number equals the number of protons. The number of electrons equals the number of protons. To find the number of neutrons subtract the atomic number from the mass number.
3. **Identify the subatomic particle used to define an element, and explain why this particle is used.** The proton defines the element's number because the each element always has the same number of protons, while the number of neutrons and electrons can vary.
4. **Explain why the masses of atoms of the same element may differ.** Because they can have different number of neutrons. For example Hydrogen (H) H, H₁Deuterium, H₂ Tritium. These are called isotopes.

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Engineering,National Science Resources Center of the National Academy of Sciences,Institute of Medicine,1998-04-30 With age appropriate inquiry centered curriculum materials and sound teaching practices middle school science can capture the interest and energy of adolescent students and expand their understanding of the world around them Resources for Teaching Middle School Science developed by the National Science Resources Center NSRC is a valuable tool for identifying and selecting effective science curriculum materials that will engage students in grades 6 through 8 The volume describes more than 400 curriculum titles that are aligned with the National Science Education Standards This completely new guide follows on the success of Resources for Teaching Elementary School Science the first in the NSRC series of annotated guides to hands on inquiry centered curriculum materials and other resources for science teachers The curriculum materials in the new guide are grouped in five chapters by scientific area Physical Science Life Science Environmental Science Earth and Space Science and Multidisciplinary and Applied Science They are also grouped by type core materials supplementary units and science activity books Each annotation of curriculum material includes a recommended grade level a description of the activities involved and of what students can be expected to learn a list of accompanying materials a reading level and ordering information The curriculum materials included in this book were selected by panels of teachers and scientists using evaluation criteria developed for the guide The criteria reflect and incorporate goals and principles of the National Science Education Standards The annotations designate the specific content standards on which these curriculum pieces focus In addition to the curriculum chapters the guide contains six chapters of diverse resources that are directly relevant to middle

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machine M2M communication applications The LPWANs are resource constrained networks and have critical requirements for long battery life extended coverage high scalability and low device and deployment costs There are several design and deployment challenges such as media access control spectrum management link optimization and adaptability energy harvesting duty cycle restrictions coexistence and interference interoperability and heterogeneity security and privacy and others LPWAN Technologies for IoT and M2M Applications is intended to provide a one stop solution for study of LPWAN technologies as it covers a broad range of topics and multidisciplinary aspects of LPWAN and IoT Primarily the book focuses on design requirements and constraints channel access spectrum management coexistence and interference issues energy efficiency technology candidates use cases of different applications in smart city healthcare and transportation systems security issues hardware software platforms challenges and future directions Addison-Wesley Introduction to Physical Science Michael B. Leyden,1988 **Light-Emitting Diodes (4th Edition, 2023)** E. Fred Schubert,2023-03-11 The 1st edition of the book Light Emitting Diodes was published in 2003 The 2nd edition was published in 2006 The 3rd edition was published in 2018 The current edition the 2023 edition is the most recent update of the book The book is a thorough discussion of LEDs particularly its semiconductor physics electrical optical material science thermal mechanical and chemical foundations The book presents many fundamental aspects of LED technology and includes an in depth discussion of white light emitting diodes LEDs phosphor materials used in white LEDs packaging technology and the various efficiencies and efficacies encountered in the context of LEDs The background of light color science and human vision is provided as well The fully colored illustrations of the current edition are beneficial given the prominent role of light and color in the field of LEDs The current edition is published in electronic PDF format in order to make the book affordable and easily accessible to a wide readership **Applied Mechanics Reviews** ,1969 **Publications of the National Bureau of Standards ... Catalog** United States. National Bureau of Standards,1979 **Mosby's Fundamentals of Therapeutic Massage - E-Book** Sandy Fritz,Luke Allen Fritz,2024-05-28 Gain the knowledge and skills you need to succeed in massage therapy Mosby s Fundamentals of Therapeutic Massage 8th Edition helps you prepare for licensing and certification exams as well as professional practice The book provides an in depth understanding of the principles of therapeutic massage and helps you develop the ability to reason effectively and make informed decisions from assessing problems and planning treatment to mastering massage protocols and techniques Hundreds of photographs demonstrate massage techniques step by step and case studies allow you to apply concepts to real world situations Featuring dozens of online how to videos this resource from massage therapy experts Sandy and Luke Fritz is your text of choice Comprehensive coverage includes all of the fundamentals of therapeutic massage including massage techniques equipment and supplies wellness working with special populations and business considerations Step by step full color photographs demonstrate massage protocols and techniques by body area Competency based approach helps you prepare for and pass licensing and certification exams including the

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 Propulsion Needs,2007-01-14 Rocket and air breathing propulsion systems are the foundation on which planning for future
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 von Karman wrote *Science the Endless Frontier* and *Toward New Horizons* respectively there has been a consensus that
 strong Department of Defense support of science and technology S T is important to the security of the United States During
 the Cold War as it faced technologically capable adversaries whose forces potentially outnumbered U S forces the United
 States relied on a strong defense S T program to support the development of technologically superior weapons and systems
 that would enable it to prevail in the event of conflict Since the end of the Cold War the United States has relied on its
 technological superiority to maintain a military advantage while at the same time reducing the size of its forces Over the past
 half century creating and maintaining a technologically superior military capability have become fundamental to U S national

security strategy and investment in S T has become a basic component of the defense budget In late 1998 Congress asked the Secretary of Defense to conduct a study in cooperation with the National Research Council NRC on the S T base of the U S Department of Defense DoD Congress was particularly concerned about areas of the S T program related to air systems space systems and supporting information systems Its concern was based on the Air Force s reduction of its S T program from the largest of the three military service programs to the smallest Congress also wanted to ensure that the Air Force maintained an appropriately sized S T workforce In late 1999 the Deputy Under Secretary of Defense for Science and Technology asked the NRC to conduct a study to explore these issues *Harcourt Science: Physical science [grade] 6, units E and F, teacher's ed ,2000* *Holt Science Spectrum Physical Science Chapter 4 Resource File: Atoms* Holt Rinehart & Winston,2008-01-01 NBS Special Publication ,1968 **Publications** United States. National Bureau of Standards,1980

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