

## Unit 1 Study Guide — ANSWER KEY

1. Define scientific method: a procedure for collecting, testing, and communicating information
2. Define hypothesis: a suggested answer to the question, an educated guess
3. Define experiment: a scientific test so see what will happen
4. Define dependent variable: the variable that you do NOT control. (on a graph is usually on the Y axis)
5. Define independent variable: the variable that I control. (on a graph is usually on the X axis)
6. Define control: the part of the experiment where everything is left the same.
7. Define observation: information collected with any of your 5 senses
8. Define density: the ratio of mass per unit volume ( $D = m/v$ )
9. Define accuracy: the measure of how close your data is to the ACTUAL or true value.
10. Define precision: a measure of how close your data (measurements) are to each other.
11. Why do scientists use SI units of measurement? So that measurements can be understood internationally (around the world).
12. List (in the correct order) the steps of the scientific method.

1. make an observation	4. test hypothesis (experiment)
2. ask a question	5. analyze data/form a conclusion
3. form a hypothesis	6. develop a theory

13. What mnemonic device is used to remember the metric system?

K H Da B d e m

14. What is the sentence (mnemonic) that will help you remember the METRIC prefixes in order?

## King Henry Died By Drinking Chocolate Milk

15. Fill in the table below describing the metric system:

	Kilo	Hecto	Deca	Base	Deci	Centi	Milli
Abbreviation	<b>K</b>	<b>H</b>	<b>Da</b>	<b>g, m, L</b>	<b>d</b>	<b>c</b>	<b>m</b>
Meaning in powers of 10	<b><math>10^3</math></b>	<b><math>10^2</math></b>	<b><math>10^1</math></b>	<b><math>10^0</math></b>	<b><math>10^{-1}</math></b>	<b><math>10^{-2}</math></b>	<b><math>10^{-3}</math></b>

**Convert the following:**

16. 25 L to 25,000 mL
17. 2.3 km to 230,000 cm
18. 0.045 g to 45 mg
19. 0.35 kg to 350 g
20. 300 mL to 0.300 L

21. 25 cm to 0.25 m  
22. 56784 mm to 0.056784 km  
23. 15 g to 0.015 kg  
24. 3.452 kL to 3452 L  
25. 47 m to 47000 mm

**Scientific Notation:**

Convert the following to scientific notation:

- |            |                       |
|------------|-----------------------|
| 26) 0.005  | $5 \times 10^{-3}$    |
| 27) 5.050  | $5.05 \times 10^{-1}$ |
| 28) 0.0008 | $8 \times 10^{-4}$    |
| 29) 1.000  | $1 \times 10^0$       |

- |            |                      |
|------------|----------------------|
| 30) 0.025  | $2.5 \times 10^{-2}$ |
| 31) 0.0025 | $2.5 \times 10^{-3}$ |
| 32) 500    | $5 \times 10^2$      |
| 33) 5,000  | $5 \times 10^3$      |

# Scientific Measurement Guide Answer Key

**ML Yell**



## Scientific Measurement Guide Answer Key:

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## **Scientific Measurement Guide Answer Key** Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Scientific Measurement Guide Answer Key**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we shall delve into the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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