

Unit 1 Study Guide — ANSWER KEY

1. Define scientific method: **a procedure for collecting, testing, and communicating information**
2. Define hypothesis: **a suggested answer to the question, an educated guess**
3. Define experiment: **a scientific test so see what will happen**
4. Define dependent variable: **the variable that you do NOT control. (on a graph is usually on the Y axis)**
5. Define independent variable: **the variable that I control. (on a graph is usually on the X axis)**
6. Define control: **the part of the experiment where everything is left the same.**
7. Define observation: **information collected with any of your 5 senses**
8. Define density: **the ratio of mass per unit volume (D = m/v)**
9. Define accuracy: **the measure of how close your data is to the ACTUAL or true value.**
10. Define precision: **a measure of how close your data (measurements) are to each other.**
11. Why do scientists use SI units of measurement? **So that measurements can be understood internationally (around the world).**
12. List (in the correct order) the steps of the scientific method.

- | | |
|--|--|
| <ol style="list-style-type: none"> 1. make an observation 2. ask a question 3. form a hypothesis | <ol style="list-style-type: none"> 4. test hypothesis (experiment) 5. analyze data/form a conclusion 6. develop a theory |
|--|--|

13. What mnemonic device is used to remember the metric system?

K H Da B d c m
g, m, L

14. What is the sentence (mnemsonic) that will help you remember the METRIC prefixes in order?

King Henry Died By Drinking Chocolate Milk

15. Fill in the table below describing the metric system:

| | Kilo | Hecto | Deca | Base | Deci | Centi | Milli |
|-------------------------|--------------------------|--------------------------|--------------------------|--------------------------|-----------------------------|-----------------------------|-----------------------------|
| Abbreviation | K | H | Da | g, m, L | d | c | m |
| Meaning in powers of 10 | 10^3 | 10^2 | 10^1 | 10^0 | 10^{-1} | 10^{-2} | 10^{-3} |

Convert the following:

- | | |
|---|---|
| <ol style="list-style-type: none"> 16. 25L to <u>25000</u> mL 17. 2.3 km to <u>230,000</u> cm 18. 0.045 g to <u>45</u> mg 19. 0.35 kg to <u>350</u> g 20. 300 mL to <u>0.300</u> L | <ol style="list-style-type: none"> 21. 25 cm to <u>0.25</u> m 22. 56784 mm to <u>0.056784</u> km 23. 15 g to <u>0.015</u> kg 24. 3.452 kL to <u>3452</u> L 25. 47 m to <u>47000</u> mm |
|---|---|

Scientific Notation:

Convert the following to scientific notation:

- | | |
|--|---|
| <ol style="list-style-type: none"> 26) 0.005 <u>5×10^{-3}</u> 27) 5.050 <u>5.05×10^1</u> 28) 0.0008 <u>8×10^{-4}</u> 29) 1,000 <u>1×10^3</u> | <ol style="list-style-type: none"> 30) 0.025 <u>2.5×10^{-2}</u> 31) 0.0025 <u>2.5×10^{-3}</u> 32) 500 <u>5×10^2</u> 33) 5,000 <u>5×10^3</u> |
|--|---|

Scientific Measurement Study Guide Answer Key

**Susan K Grove, Assistant Professor of
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Fcn Faan, Jennifer Gray**

Scientific Measurement Study Guide Answer Key:

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Table of Contents Scientific Measurement Study Guide Answer Key

1. Understanding the eBook Scientific Measurement Study Guide Answer Key
 - The Rise of Digital Reading Scientific Measurement Study Guide Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Scientific Measurement Study Guide Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Scientific Measurement Study Guide Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Scientific Measurement Study Guide Answer Key
 - Personalized Recommendations
 - Scientific Measurement Study Guide Answer Key User Reviews and Ratings
 - Scientific Measurement Study Guide Answer Key and Bestseller Lists

5. Accessing Scientific Measurement Study Guide Answer Key Free and Paid eBooks
 - Scientific Measurement Study Guide Answer Key Public Domain eBooks
 - Scientific Measurement Study Guide Answer Key eBook Subscription Services
 - Scientific Measurement Study Guide Answer Key Budget-Friendly Options
6. Navigating Scientific Measurement Study Guide Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Scientific Measurement Study Guide Answer Key Compatibility with Devices
 - Scientific Measurement Study Guide Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Scientific Measurement Study Guide Answer Key
 - Highlighting and Note-Taking Scientific Measurement Study Guide Answer Key
 - Interactive Elements Scientific Measurement Study Guide Answer Key
8. Staying Engaged with Scientific Measurement Study Guide Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Scientific Measurement Study Guide Answer Key
9. Balancing eBooks and Physical Books Scientific Measurement Study Guide Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Scientific Measurement Study Guide Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Scientific Measurement Study Guide Answer Key
 - Setting Reading Goals Scientific Measurement Study Guide Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Scientific Measurement Study Guide Answer Key
 - Fact-Checking eBook Content of Scientific Measurement Study Guide Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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