

Chemistry Practice Writing Chemical Formulas

Write a chemical formula for each compound.

1.	<u>Al_2O_3</u>	aluminum oxide
2.	<u>H_2SO_4</u>	sulfuric acid
3.	<u>H_2S</u>	hydrogen sulfide
4.	<u>K_2SO_4</u>	potassium sulfate
5.	<u>H_2CO_3</u>	carbonic acid
6.	<u>H_2SO_3</u>	sulfurous acid
7.	<u>$\text{Ca}_3(\text{PO}_4)_2$</u>	calcium phosphate
8.	<u>H_2CO_3</u>	carbonic acid
9.	<u>$\text{Ca}_3(\text{PO}_4)_2$</u>	calcium phosphate
10.	<u>H_2SO_4</u>	sulfuric acid
11.	<u>H_2SO_4</u>	sulfuric acid
12.	<u>H_2SO_4</u>	sulfuric acid
13.	<u>H_2SO_4</u>	sulfuric acid
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78.	<u>H_2SO_4</u>	sulfuric acid
79.	<u>H_2SO_4</u>	sulfuric acid
80.	<u>H_2SO_4</u>	sulfuric acid

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Jessica J Manson



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