

Physical Science Chapter 2 Test**Completion**

Completion: write complete or statements.

1. Based on our model of matter because of how we _____.
2. Atoms, ions, molecules, and compounds are composition of _____.
3. The chemical symbol of oxygen for the element _____.
4. A substance that is made of atoms of same kind are kept bound together is called a _____.
5. Every chemical compound is unique and different from the _____ in nature.
6. The smallest particle of oxygen that behaves like oxygen is an oxygen _____.
7. The chemical formula for a chlorine molecule, which consists two chlorine atoms, is _____.
8. Each molecule of ammonia, NH_3 , contains _____ atoms of ions.
9. Each molecule of hydrochloric acid HCl contains _____ hydrogen atoms.
10. When two or more pure substances are mixed together, the result is a _____.
11. Two or more liquids that can dissolve into each other are referred to as _____.
12. A mixture of dust and water is called a _____ mixture because the substances do not dissolve.
13. The kinetic theory states that the particles in matter are always in _____.
14. The kinetic theory states that, at the same _____, heavier particles move more slowly than lighter particles.
15. As gas molecules travel around each other, they spread in _____.
16. A solid holds its shape because its molecules are _____.
17. As heat is added to a solid substance, the atoms _____ and move apart.
18. When water _____, steam is released.
19. One way change of state is called _____, which is evaporation.
20. The surface of the liquid has several smaller air bubbles smaller because of _____.
21. Energy can be converted from one form to another, but it cannot be _____.
22. When you burn gasoline in a car's engine, you produce energy to move the car plus energy in the exhaust _____.
23. When a piece of paper is burned, the amount of matter before is _____ the amount of matter afterward.

Physical Science Chapter 2 Test

Caren Van Slyke



Physical Science Chapter 2 Test:

Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science ,2003-11

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