

CHAPTER 7

Half-Life and Radioisotope Dating

BLM 2-48

Answer the questions in the space provided.

1. Examine the graph showing the decay curve for carbon-14. The graph shows the amount of radioactive carbon-14 that would be in a sample of organic material for 30 000 years after the organism died.



- (a) Define half-life.
- (b) How long is one half-life for carbon-14? _____
- (c) What percentage of carbon-14 remains (i) after one half-life? _____
 (ii) after two half-lives? _____ (iii) after three half-lives? _____
- (d) Use the graph to estimate the percentage of carbon-14 remaining after
 (i) 5000 years _____ (ii) 10 000 years _____
 (iii) 15 000 years _____
- (e) Use the graph to estimate the number of years that have passed since the organism died if the percentage of parent isotope that remains is
 (i) 40% _____ (ii) 20% _____ (iii) 5% _____
- (f) Explain why carbon-14 half-life measurements are not effective in dating an organism that has been dead for more than 50 000 years.

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KJ Lindholm-Leary



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Table of Contents Physical Science If8767 Pg 100 Answers Half Life

1. Understanding the eBook Physical Science If8767 Pg 100 Answers Half Life
 - The Rise of Digital Reading Physical Science If8767 Pg 100 Answers Half Life
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science If8767 Pg 100 Answers Half Life
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science If8767 Pg 100 Answers Half Life
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science If8767 Pg 100 Answers Half Life
 - Personalized Recommendations
 - Physical Science If8767 Pg 100 Answers Half Life User Reviews and Ratings
 - Physical Science If8767 Pg 100 Answers Half Life and Bestseller Lists

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 - Physical Science If8767 Pg 100 Answers Half Life Public Domain eBooks
 - Physical Science If8767 Pg 100 Answers Half Life eBook Subscription Services
 - Physical Science If8767 Pg 100 Answers Half Life Budget-Friendly Options
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 - ePub, PDF, MOBI, and More
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7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science If8767 Pg 100 Answers Half Life
 - Highlighting and Note-Taking Physical Science If8767 Pg 100 Answers Half Life
 - Interactive Elements Physical Science If8767 Pg 100 Answers Half Life
8. Staying Engaged with Physical Science If8767 Pg 100 Answers Half Life
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science If8767 Pg 100 Answers Half Life
9. Balancing eBooks and Physical Books Physical Science If8767 Pg 100 Answers Half Life
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science If8767 Pg 100 Answers Half Life
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Physical Science If8767 Pg 100 Answers Half Life
 - Setting Reading Goals Physical Science If8767 Pg 100 Answers Half Life
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Science If8767 Pg 100 Answers Half Life
 - Fact-Checking eBook Content of Physical Science If8767 Pg 100 Answers Half Life
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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