Section 5.2 The Modern Periodic Table (pages 130-138)

This section explains the organization of the modern periodic table and discusses the general properties of metals, nonmetals, and metalloids.

Reading Strategy (page 130)

Previewing Before you read, complete the table by writing two questions about the periodic table on pages 132-133. As you read, write answers to your questions. For more information on this reading strategy, see the Reading and Study Skills in the Skills and Reference Handbook at the end of your textbook.

Que	estions About the Periodic Table	
Question	Answer	

The Periodic Law (pages 131-133)

- 1. Is the following sentence true or false? In the modern periodic table, elements are arranged by increasing number of protons.
- Explain why the number of elements per period varies. ____
- 3. Properties of elements repeat in a predictable way when atomic numbers are used to arrange elements into groups. This pattern of repeating properties is called the

Atomic Mass (page 134)

4. Label the four types of information supplied for chlorine in the diagram.



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Douglas Fisher, Nancy Frey, John Hattie

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communication modules on degree programmes to vocational courses in health business and education With contributions from an internationally renowned range of scholars this is the definitive text for students researchers and professionals alike

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available when I started my clinical placements Second year medical student University of Liverpool It helps aid students to learn effectively and efficiently and even tells you how you will know when you know enough Professor Parveen Kumar

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TRAM 11, Training Resources Applied to Mining Michael J. Klishis, Ann Payne, 1985

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