

## Gas Laws Computer Lab

### Purpose

- 1.) To identify and describe the 4 properties of a gas sample: moles ( $n$ ), volume ( $V$ ), temperature ( $T$ ), pressure ( $P$ )
- 2.) To determine the relationship between 2 variables at a time while keeping the other 2 variables constant
  - a.  $P$  and  $V$  (while keeping  $n$  and  $T$  constant)
  - b.  $V$  and  $T$  (while keeping  $n$  and  $P$  constant)
  - c.  $P$  and  $T$  (while keeping  $n$  and  $V$  constant)
  - d.  $V$  and  $n$  (while keeping  $P$  and  $T$  constant)

### Definition of gas variables

A sample of gas in a closed container has 4 main characteristics:

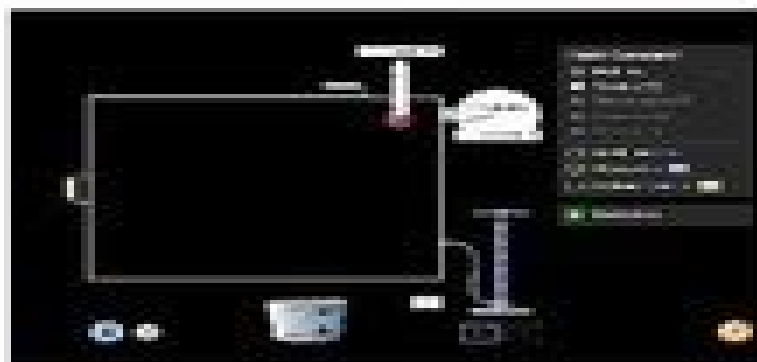
- a. Number of \_\_\_\_\_ of gas ( $n$ ) – the quantity of gas particles present
- b. Volume of gas ( $V$ ) – the amount of \_\_\_\_\_ the gas sample occupies
- c. \_\_\_\_\_ of gas ( $T$ ) – the average kinetic energy of the gas particles (i.e. how fast they're moving)
- d. Pressure of gas ( $P$ ) – the frequency and forcefulness of \_\_\_\_\_ between particles and container walls

### Procedure

#### Part A – Set Up

- 1.) Google "phet gases intro", click the first search result, click PLAX button, double-click "Gases", or go to [https://phet.colorado.edu/en/simulations/where-to-play/gases-intro\\_en.html](https://phet.colorado.edu/en/simulations/where-to-play/gases-intro_en.html)

Your screen should look like this:



- 2.) Open the "Particles" box. Make sure the like pump is purple color for heavy particles. Move the like pump handle to the highest position and push down to pump gas particles into the container. There are now 30 heavy gas particles in the container. In the "Particles" box, use the arrow buttons to make 100 heavy gas particles in container.

# Phet Gas Law Simulation Lab Answers

**Tai, Chih-Che, Moran, Renee M.  
R., Robertson, Laura, Keith,  
Karin, Hong, Huili**



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