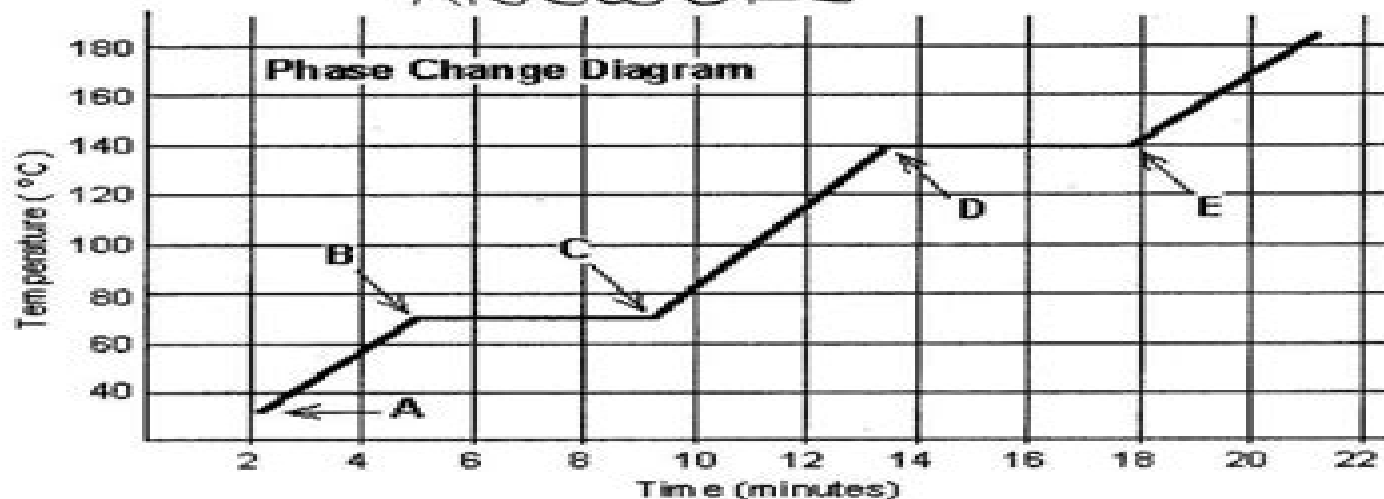


Phase Change Worksheet II

ANSWERS



The graph was drawn from data collected as a substance was heated at a constant rate. Use the graph to answer the following questions.

At point A, the beginning of observations, the substance exists in a solid state. Material in this phase has DEFINITE volume and DEFINITE shape. With each passing minute, HEAT is added to the substance. This causes the molecules of the substance to MOVE/IRRATE more rapidly which we detect by a TEMPERATURE rise in the substance. At point B, the temperature of the substance is 70 °C. The solid begins to MELT. At point C, the substance is completely MELTED or in a LIQUID state. Material in this phase has DEFINITE volume and INDEFIN shape. The energy put to the substance between minutes 5 and 9 was used to convert the substance from a SOLID to a LIQUID.

Between 9 and 13 minutes, the added energy increases the TEMP of the substance. During the time from point D to point E, the liquid is BOILING. By point E, the substance is completely in the GAS phase. Material in this phase has INDEFIN volume and INDEFIN shape. The energy put to the substance between minutes 13 and 18 converted the substance from a LIQUID to a GAS state. Beyond point E, the substance is still in the GAS phase, but the molecules are moving FASTER/MORE as indicated by the increasing temperature.

Phase Change Diagram Answer Key

Carson-Dellosa Publishing

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the Carson-Dellosa Publishing text.

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