

Gases Unit Virtual Lab: Balloons and Buoyancy in Air

Question: How do gases in flexible and rigid containers achieve and lose buoyancy?

Learning Goals: Students will (on a molecular level) be able to...

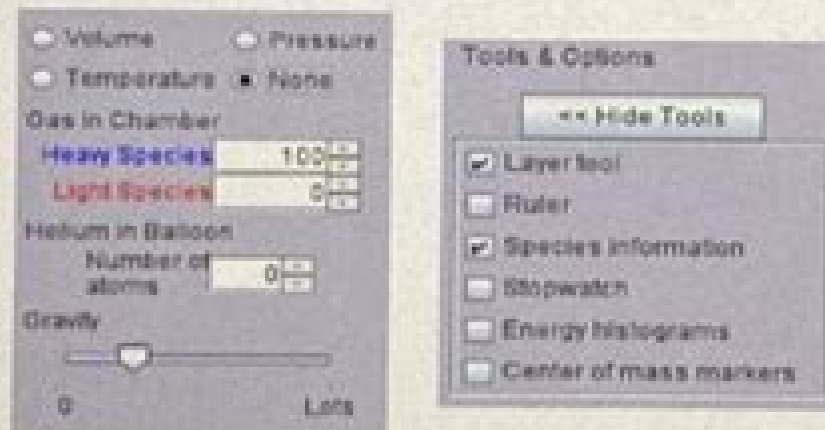
- 1) Determine what causes a helium balloon to rise or sink in the air around it.
- 2) Explain why a rigid sphere containing helium floats or sinks.
- 3) Explain why a hot air balloon must have a heater to keep it aloft (floating in air).
- 4) Describe the differences between how a hot air balloon, rigid sphere, and helium balloon function and why.

Directions:

- Don't vary the gravity for this activity.
- Use the procedure provided under each learning goal to conduct your investigation.
- Describe in writing your observations as you progress through the steps. After completing the procedure, write in paragraph form an explanation for what you observed based on your knowledge of the kinetic molecular theory and the relationships between gas variables (pressure, volume, temperature, kinetic energy, velocity, and number of gas particles). Refer to your data and observations to support your explanation.

Goal 1: Determine what causes a helium balloon surrounded by air to rise or sink.

1. Open the PhET simulation "Balloons & Buoyancy". Choose the "Helium Balloon" setting from the top menu bar. Set the initial conditions to match those shown in the diagram below.



Phet Lab Balloons And Buoyancy Virtual Lab Answers

Mark Steyvers



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