

## SDSU CHEMISTRY PLACEMENT TEST WITH COMPLETE SOLUTIONS 100% 2023/2024

proton  
-heavy  
-positively charged  
-nucleon

neutron  
-heavy  
-neutral  
-nucleon

electron  
-very small (1/1800 of proton)  
-negative  
-outside of nucleus

properties of nucleus  
-# of protons determines the element  
-held together by nuclear force  
-# of protons is extremely stable unlike # of electrons

atomic number  
equal to the # of protons

atomic mass  
# of protons + # of neutrons averaged across all of the element on earth

isotopes  
-when elements have different masses but the same chemical properties  
-ex. silver-107 and silver-109

lewis dot  
-a 2-D model that represents covalent bonds as straight lines, ionic bonds as spaces, and unbonded valence electrons as dots  
-can have single, double, or triple bonds

physical properties of compounds and molecules  
-melting point  
-boiling point  
-solubility  
-electrical conductivity

molecular mass  
# of atoms(atomic mass)=molecular mass

# Sdsu Chemistry Placement Test Study Guide

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A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the National Learning Corporation text.

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