

# Electrons in Atoms

## Section 5.1 Light and Quantized Energy

In your textbook, read about the wave nature of light.

Use each of the terms below just once to complete the passage.

amplitude  
light

energy  
wave

frequency  
wavelength

hertz  
speed

Electromagnetic radiation is a kind of **(1)** \_\_\_\_\_ that behaves like a(n) **(2)** \_\_\_\_\_ as it travels through space. **(3)** \_\_\_\_\_ is one type of electromagnetic radiation. Other examples include X rays, radio waves, and microwaves.

All waves can be characterized by their wavelength, amplitude, frequency, and **(4)** \_\_\_\_\_. The shortest distance between equivalent points on a continuous wave is called a(n) **(5)** \_\_\_\_\_. The height of a wave from the origin to a crest or from the origin to a trough is the **(6)** \_\_\_\_\_. **(7)** \_\_\_\_\_ is the number of waves that pass a given point in one second. The SI unit for frequency is the **(8)** \_\_\_\_\_, which is equivalent to one wave per second.

Use the figure to answer the following questions.



9. Which letter(s) represent one wavelength? \_\_\_\_\_
10. Which letter(s) represent the amplitude? \_\_\_\_\_
11. If twice the length of A passes a stationary point every second, what is the frequency of the wave?  
\_\_\_\_\_

# Prentice Hall Chemistry Chapter 5 Notes

**Michael Grossman**



## **Prentice Hall Chemistry Chapter 5 Notes:**

Prentice Hall Science Explorer Michael J. Padilla, 2002      Forensic Chemistry Michael Grossman, 2021-12-20 FORENSIC CHEMISTRY FUNDAMENTALS strives to help scientists workplace health safety and other areas are included And two issues of Canadian legal process are argued as essays in the final two chapters      Introduction to Computational Chemistry Frank Jensen, 2013-03-22 Introduction to Computational Chemistry Second Edition provides a comprehensive account of the fundamental principles underlying different methods ranging from classical to the sophisticated Although comprehensive in its coverage this textbook focuses on calculating molecular structures and relative energies and less on molecular properties or dynamical aspects No prior knowledge of concepts specific to computational chemistry are assumed but the reader will need some understanding of introductory quantum mechanics linear algebra and vector differential and integral calculus

**Fundamental Principles of International Relations** J. Martin Rochester, 2018-04-17 This book distills the essential elements of world politics both the enduring characteristics as well as the revolutionary changes that may be altering the very fabric of the centuries old state system Author J Martin Rochester explores all the important topics that one would expect to find in an IR text war diplomacy foreign policy international law and organization the international economy and more but injects fresh perspectives on how globalization and other contemporary trends are affecting these issues In addition the author does so through a highly engaging lively writing style that will appeal to today's students Fundamental Principles of International Relations is a tightly woven treatment of international politics past and present drawing on the latest academic scholarship while avoiding excessive jargon and utilizing pedagogical aids while avoiding clutter Rochester ultimately challenges the reader to think critically about the future of a post Cold War and post 9 11 world that is arguably more complex if not more dangerous than some previous eras with the potential for promise as well as peril      Kinetics of Geochemical Processes Anthonio C. Lasaga, James Kirkpatrick, 2018-12-17 Volume 8 of Reviews in Mineralogy treats a Short Course in Kinetics which brings together the fundamentals needed to explain field observations using kinetic data It is hoped that this book may serve not only as a reference for researchers dealing with the rates of geochemical processes but also as a text in courses on geochemical kinetics The book is organized with a rough temperature gradient in mind i e low temperature kinetics at the beginning and igneous kinetics at the end However the topics in each chapter are general enough that they can be applied often to any geochemical domain sedimentary metamorphic or igneous The theory of kinetics operates at two complementary levels the phenomenological and the atomistic The former relies on macroscopic variables e g temperature or concentrations to describe the rates of reactions or the rates of transport the latter relates the rates to the basic forces operating between the particular atomic or molecular species of any system This book deals with both descriptions of the kinetics of geochemical processes      **Quantitative Chemical Analysis** Daniel C. Harris, 2010-04-30 QCA is the bestselling textbook of choice for analytical chemistry It offers a modern portrait of the techniques of chemical analysis backed by a

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