Name KEY Pr. Date Write balanced formula equations for these reactions and identify the type of reaction. ►2KI Potassium + Iodine Synthesis __ 2K + I2 Barium Iodide + Sodium Sulfate ______ double rep BaIz + Na 2 SO4 -> Ba SO4 + 2NaI Aluminum Hydroxide decomposition 2 AI(OH)3 -> AI2O3 + 3H2O 4. Sodium Chlorate NaCl + Oz decomposition Aluminum Bromide + Chlorine ---single replacement AIBry + Cl2 -> AICI3+ Bry 6. Phosphoric Acid + Magnesium ----H3PO4 + Mg -> Mq3(PO4)2+ H2 7. Iron III Sulfate + Calcium Nitrate ---double rep. Fez(Say)3 +3Ca(NO3)2 -> 2Fe(NO3)2+3Ca SO4 8. Sodium + Water w Stragte vep ZNa+2H2O -> ZNaOH + HZ 9. Aluminum + Iodine ---synthesis 2 AL +3T2 -> ZALT2 10. Iron III Hydroxide + Hydrobromic Acid double vep. Fe (OH), +3HBr -> FeBr, +3HOH Magnesium Hydroxide + Ammonium Sulfate — double rep Mg(OH). + (NH4) SO4 -> MgSG4 + 2NH4OH 12. Water -12H2+ Dx decomposition 2420 13. Calcium + Hydrochloric Acid ---single repassment ca +2HCI -> Cacis + Hz 14. Lithium + Chlorine -Synthesis 2 Li+Ch -> 2 LiC1

Predict Balancing Equations Answers

Olga I. Agapova

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The Sourcebook for Teaching Science, Grades 6-12 Norman Herr,2008-08-11 The Sourcebook for Teaching Science is a unique comprehensive resource designed to give middle and high school science teachers a wealth of information that will enhance any science curriculum Filled with innovative tools dynamic activities and practical lesson plans that are grounded in theory research and national standards the book offers both new and experienced science teachers powerful strategies and original ideas that will enhance the teaching of physics chemistry biology and the earth and space sciences

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