

Chemistry Chapter 10 Test Practice Problems

Convert each of the following to **moles**:

- a. 22.0 g NO_2
- b. 190 g MgF_2
- c. 2.35×10^{25} atom K
- d. 18 molecules Na_2O

Convert each of the following to **grams**:

- a. 4.10 mol CaCO_3
- b. 10.2 mol C_2P_3
- c. 3.01×10^{23} atoms Li
- d. 1.2×10^{18} molecules $\text{C}_3\text{H}_{10}\text{O}_5$

Convert each of the following to **liters**:

- a. 1.567 mol Ne
- b. 3.56×10^{27} atoms of He
- c. 6.0 g O_2
- d. 2.51×10^{20} molecules CO_2

Calculate the **percentage composition** for each element in the following compounds:

- | | |
|--------------------|----------------------------|
| a. MgCl_2 | c. Pb_2O_3 |
| b. CaI_2 | d. AuSO_4 |

Prentice Hall Chemistry Answer Key Chemical Quantities

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Decoding **Prentice Hall Chemistry Answer Key Chemical Quantities**: Revealing the Captivating Potential of Verbal Expression

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