

UNIT 2 2015 TO 2021 PAPER 2

HL IB CHEMISTRY *[67 marks]*

Iron may be extracted from iron (II) sulfide, FeS.

Iron (II) sulfide, FeS, is ionically bonded.

1. State the full electron configuration of the sulfide ion. *[1 mark]*

.....
.....
.....

Markscheme

$1s^2 2s^2 2p^6 3s^2 3p^6$ ✓

*Do **not** accept "[Ne] 3s² 3p⁶".*

Magnetite, Fe₃O₄, is another ore of iron that contains both Fe²⁺ and Fe³⁺.

Iron exists as several isotopes.

- 2a. State the type of spectroscopy that could be used to determine their relative abundances. *[1 mark]*

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Paper 2 Ib Chemistry Hl

Julia Schneider



Paper 2 Ib Chemistry HL:

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