

Workbook ANSWER KEY

1 FRIENDS AND FAMILY

LESSON A • What kind of person are you?

1 Grammar

- a. 1 c. 1 e. 1 g. 1 i. 1 k. 1, 2
b. 1 d. 1, 2 f. 1 h. 1, 2 j. 1, 2 l. 1

2 Grammar

1. Sam can't stand having / to have nothing to do.
2. Vic and Joon avoid going to school parties.
3. Leo prefers visiting / to visit his parents on Sundays / the weekends.
4. Ivy is into learning new languages.
5. Sue is worried about going rock climbing.
6. Celia enjoys reading to kids / volunteering as a storyteller.

3 Grammar

Answers will vary.

4 Vocabulary

- A** 1. b 4. f 7. h
2. a 5. e 8. i
3. c 6. d 9. g

B Answers will vary.

5 Writing

- A** 1. My mother is very adventurous.
2. My friend John always says what is on his mind.

B Answers will vary.

LESSON B • Every family is different.

1 Grammar

I love my family so much, and I really get along with everyone – my parents and my four brothers and sisters. However, sometimes they drive me crazy. There are both good and bad things about coming from a large family. One of the best things about coming from a large family is that I always have someone to talk to. Unfortunately, one of the disadvantages is that I never have any privacy. And of course, the trouble with not having any privacy is that I never have any space I can call my own. Our house is big, but sometimes not big enough!

2 Grammar

1. The best thing about being the youngest is that I'm the center of attention.
2. The disadvantage of having a lot of kids is that I can't give each of them the individual attention they want.
3. The problem with living with my father-in-law is that we disagree about everything.
4. The worst thing about having two younger sisters is that they always want to know all about my personal life.
5. The trouble with having an identical twin is that no one can ever tell us apart.

3 Grammar

Answers will vary.

4 Vocabulary

1. False 3. True 5. False 7. True 9. False
2. True 4. False 6. True 8. True 10. True

5 Reading

- A** 1. c 2. b 3. a

- B** 1. True
2. False; When only children grow up, they are just as sociable as children with siblings.
3. False; Only children score better on tests, but so do first-born children and those with one sibling.
4. True

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A red circular graphic with a gradient, appearing as a partial circle or a stylized 'C' shape, located to the right of the American Mathematical Society text.

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the boiling point of an organic liquid is the temperature at which its vapour pressure equals the atmospheric pressure over the liquid or it is the temperature at which the vapour and liquid phases are in equilibrium at a given pressure the boiling point is considered as a criterion of purity of a compound and is useful for identification of

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given that the boiling points of ethanol and methanol 78 c and 64 c respectively are far off in a significant way this seems like a valid conclusion conclusion thermometer calibrations are necessary in order to collect reliable information as precise calculations are imperative to gathering correct results

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melting point 1 a push the open end of a capillary tube into the powdered acetamide b move the powder to the closed end of the capillary tube by tapping it on the table repeat until the powdered acetamide occupies 1 2 mm of the capillary tube end c

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conclusion after completing this experiment the melting point of three different unknown compounds was obtained using the mel temp apparatus compound a had a melting point range of 114 6 115 5 c thus revealing it to be acetanilide compound b had a melting point range of 122 6 124 5 c thus

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melting point can be used as the identity and purity of an organic solid however it is not the same as the boiling point the boiling point is a characteristic unique of pure liquids the boiling point is the temperature at which the vapor pressure of a liquid is equal to the atmospheric pressure

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aug 11 2020 boiling points are very sensitive to changes in applied pressure so all boiling points should be reported with the measured pressure a compound's normal boiling point refers to its boiling point at a pressure of 760 mm Hg

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the purpose of the experiment is to understand what vapour pressure is and how it changes with temperature to understand the relationship between vapour pressure at a given temperature and boiling temperature at a given pressure to determine the vapour pressure of a pure liquid at various temperatures

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laboratory partners samantha and mackenzie goals □ correctly determine the boiling point of an unknown liquid □ measure the refractive index of an unknown □ use the gas chromatogram to determine the mass percentage of a mixture and

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50 °C so in order to convert an experimental measurement taken in Calgary which is at higher altitude than sea level and so is at a lower pressure to that reported for sea level higher pressure one needs to add a correction factor since the boiling point at sea level is higher than that at higher altitudes

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there are a variety of methods by which a sample's boiling point can be determined including distillation reflux and by using a Thiele tube the most straightforward method uses a Thiele tube and has the advantage of using less than 0.5 mL of material

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