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#### FIRST QUARTER EXAMINATION PHYSICAL SCIENCE

| Name: _ | Name:   |                  |                   | Date:                       |                       |
|---------|---|------------------|-------------------|-----------------------------|-----------------------|
| Grade L |   |                  |                   |                             |                       |
| I. MUL  | TIPLE CH  | IOICE: Write     | the letter of the | correct answer before th    | e number.             |
| 1       | Which of the following is the smallest and basic unit of matter   |                  |                   |                             |                       |
| а       | Cell  | b. Atom          | c. Molecule       | d Nucleus                   |                       |
| 2       | 2. What do you call to the combination of atoms.  |                  |                   |                             |                       |
| а       | Cell  | b. Atom          | c. Molecule       | d. Nucleus                  |                       |
| 3       | It refers to anything that occupies space and has mass.   |                  |                   |                             |                       |
| а       | Atom  | b. Matter        | c. Molecule       | d Nucleus                   |                       |
| 4       | He was an English scientist who is well known for his work in the development of the atomic theory            |                  |                   |                             |                       |
| a       | JJ Thompson b. Eugene Goldstein c. Ernest Rutherford d. John Dalton   |                  |                   |                             |                       |
| 5       | State of matter that is highly compressible because the distance of the molecules is far greater than liquid. |                  |                   |                             |                       |
| a       | Solid   | b. Liquid        | c. Gas            | d. Plasma                   |                       |
| 6       | It refers to the combination of two or more elements.   |                  |                   |                             |                       |
| а       | a. Homogeneous b. Heterogeneous c. Mixture d. Compound  |                  |                   |                             | und                   |
| 7       | . It refers to the change in its physical properties without modifying the chemical composition of<br>matter  |                  |                   |                             |                       |
| а       | Chemic  | al Change b. P   | hysical Change    | c. Organic Compound         | d. Inorganic Compound |
| 8       | t It refers t   | o the alternatio | n of matter's che | emical composition.         |                       |
| a       | Chemic  | al Change b. P   | hysical Change    | c. Organic Compound         | d Inorganic Compound  |
| 9       | . It is comp  | posed of a parti | icle, which canno | ot be separated by physical | means.                |
| a       | Element   | ь. с             | compound          | c. Mixture                  | d. Pure Substance     |
| 1       | O. What is  | the simplest fo  | em of pure subst  | ance Example Oxygen, H      | ydrogen, and Carbon   |
| a       | Element   | b. C             | compound          | c. Motture                  | d. Pure Substance     |

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## **Table of Contents Plato End Of Course Physical Science Exam**

- 1. Understanding the eBook Plato End Of Course Physical Science Exam
  - The Rise of Digital Reading Plato End Of Course Physical Science Exam
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Plato End Of Course Physical Science Exam
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - o Features to Look for in an Plato End Of Course Physical Science Exam
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Plato End Of Course Physical Science Exam
  - Personalized Recommendations
  - Plato End Of Course Physical Science Exam User Reviews and Ratings
  - Plato End Of Course Physical Science Exam and Bestseller Lists
- 5. Accessing Plato End Of Course Physical Science Exam Free and Paid eBooks
  - Plato End Of Course Physical Science Exam Public Domain eBooks
  - Plato End Of Course Physical Science Exam eBook Subscription Services
  - Plato End Of Course Physical Science Exam Budget-Friendly Options
- 6. Navigating Plato End Of Course Physical Science Exam eBook Formats

- o ePub, PDF, MOBI, and More
- Plato End Of Course Physical Science Exam Compatibility with Devices
- Plato End Of Course Physical Science Exam Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - o Adjustable Fonts and Text Sizes of Plato End Of Course Physical Science Exam
  - Highlighting and Note-Taking Plato End Of Course Physical Science Exam
  - Interactive Elements Plato End Of Course Physical Science Exam
- 8. Staying Engaged with Plato End Of Course Physical Science Exam
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Plato End Of Course Physical Science Exam
- 9. Balancing eBooks and Physical Books Plato End Of Course Physical Science Exam
  - Benefits of a Digital Library
  - o Creating a Diverse Reading Collection Plato End Of Course Physical Science Exam
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Plato End Of Course Physical Science Exam
  - Setting Reading Goals Plato End Of Course Physical Science Exam
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Plato End Of Course Physical Science Exam
  - Fact-Checking eBook Content of Plato End Of Course Physical Science Exam
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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