

Seat No.	
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T.Y. B.Sc. (Semester – III) Examination, 2012
CHEMISTRY (Paper – III)
CH – 333 : Organic Chemistry (2008 Pattern)

Time : 2 Hours

Max. Marks : 40

- N.B. :** i) All questions are **compulsory**.
 ii) Figures to the **right** indicate **full marks**.
 iii) **Draw** the structures and **neat** diagram if necessary.

1. Answer the following :

10

- i) Why urea is neutral though it contains $-\text{NH}_2$ group ?
- ii) Write the trivial and IUPAC name for $\text{ph}-\text{CH}=\text{CH}-\overset{\text{O}}{\parallel}{\text{C}}-\text{ph}$.
- iii) Draw Zig-Zag structure of hepta, 3-5-dione.
- iv) Arrange the following in decreasing order of $\text{S}_{\text{N}}2$ reactivity. Explain your reasoning.



- v) Guanidine is extremely strong base-explain in brief.
- vi) 2-phenyl ethyl bromide on heating with NaOCH_3 gives 95 % styrene.
- vii) Write the reaction of Ethyl thiol with benzaldehyde.
- viii) 1-Butene treated with HBr in presence of peroxide. Write the product.
- ix) What is $\text{S}_{\text{N}}1$ reaction ?
- x) Salicylic acid is stronger acid than m-hydroxy benzoic acid - why ?

P.T.O.

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