

Unit 5 – Macromolecules Molecular Compounds

Properties

Non-ionic/non-polar covalent

10. Name each of the following binary compounds of non-metallic elements.

a. PBr_3

b. Cl_2PBr_3

c. PCl_3

d. PBr_3

e. Cl_2PBr_3

f. PCl_3

g. Cl_2PBr_3

h. Cl_2PBr_3

i. Cl_2PBr_3

j. Cl_2PBr_3

k. Cl_2PBr_3

l. Cl_2PBr_3

m. Cl_2PBr_3

n. Cl_2PBr_3

o. Cl_2PBr_3

p. Cl_2PBr_3

q. Cl_2PBr_3

r. Cl_2PBr_3

s. Cl_2PBr_3

t. Cl_2PBr_3

11. Write the formula for each of the following binary compounds of non-metallic elements.

a. phosphorus trichloride

b. carbon tetrachloride

c. dinitrogen pentoxide

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