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P525/I CHEMISTRY (Theory) Paper I Nov. /Dec. 2019. 23/4 bours.



UGANDA NATIONAL EXAMINATIONS BOARD

Uganda Advanced Certificate of Education

(THEORY)
Paper 1

2 hours 45 minutes

INSTRUCTIONS TO CANDIDATES:

Answer all questions in Section A and six questions in Section B.

All questions must be answered in the spaces provided.

The Periodic Table, with relative atomic masses, is supplied.

Mathematical tables (3-figure tables) are adequate or non-programmable scientific electronic calculators may be used.

Illustrate your answers with equations where applicable.

Where necessary use the following:

Molar gas constant, $R = 8.31 \text{ JK}^{-1} \text{mol}^{-1}$

Molar volume of gas at s.t.p. is 22.4 litres

Standard temperature = 273 K

Standard pressure = 101325 Nm⁻²

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1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	Total

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