

Service Manual

Digital Camera

LUMIX



Model No. **DMC-TZ5P**
DMC-TZ5PC
DMC-TZ5PL
DMC-TZ5PR
DMC-TZ5EB
DMC-TZ5EE
DMC-TZ5EF
DMC-TZ5EG
DMC-TZ5E
DMC-TZ5SG
DMC-TZ15GC
DMC-TZ15GD
DMC-TZ15GK
DMC-TZ15GN
DMC-TZ15GT
DMC-TZ15GJ

VOL. 1

Colours

(S)..... Silver Type (except PR/GD)
(K)..... Black Type
(A)..... Blue Type (only DMC-TZ5P/PG/PL/PR/
EE/EG/E)
(TA)..... Dark Brown Type (only DMC-TZ5EF/
EG/E)
(T)..... Brown Type (only DMC-TZ5SG, DMC-
TZ15GC/GK/GJ)

Panasonic®

Tz5 Manual Service

Scott C. Dulebohn



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web objective to apply stoichiometric principles to calculate the appropriate chemical equation for the reaction between iron metal and a solution of copper i sulfate introduction in the form of finely split powder copper metal is produced as a byproduct of the process as iron ions the ion meta dissolves

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web purpose to enhance the understanding of stoichiometry a reaction between iron and copper ii sulfate solution will be conducted this will help you to differentiate limiting and excess reactant in a chemical reaction finally the theoretical and percent yield of this reaction will be calculated

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web purpose in this experiment you will observe a single replacement reaction you will also use stoichiometry to predict what the theoretical yield of product is and calculate a percent yield reaction iron copper ii sulfate copper iron ii sulfate pre lab questions write the balanced equation

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cylinder and add it to the beaker 4

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web under the finalize of the experiment it was found that iron was the limiting reactant and iron ii sulfate was a product of the reaction between 7 0002g away cooper ii sulfate and 2 0101g iron metal which produced 2 3037g of copper product with a percent yield of 100 72 introduction

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