

# CHEMISTRY 1

## FINAL EXAM REVIEW

### Table of Contents

30	Legal Stuff
40	Writing and Balancing Equations
50	Writing Compound Names and Formulas
60	Moles and Avogadro's Number
70	Stoichiometry
80	Empirical and Molecular Formula
90	The Periodic Table
100	Gas Law Calculations
110	Molarity and Dilution
120	Dimensional Analysis and Unit Conversions
130	Drawing Lewis Structures
140	Phase Diagrams and Phase Transitions
150	Electron Configurations
160	Intermolecular Forces
170	Periodic Trends
180	Significant Figures
190	Scientific Notation
200	Thermochemistry: Calorimetry and Enthalpy
210	Acids, Bases, and pH
220	Electrons and Quantum Numbers
230	Classifying matter, phase changes, and properties
240	Types of Chemical Reactions, Activity Series, and Solubility
25-440	Answers

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### Chemistry Review: Writing and Balancing Equations

**Directions:** Solve each problem and answer each question

1. Balance each equation below by adding the correct coefficients



2. Write and balance each equation. Include phase labels

a. Aqueous sodium hydroxide reacts with aqueous sulfuric acid to form aqueous sodium sulfate and water

b. Solid iron (III) hydroxide and aqueous sodium carbonate are formed from the reaction of solid iron (III) carbonate and aqueous sodium hydroxide

c. Solid aluminum reacts with aqueous hydrobromic acid to form aqueous aluminum bromide and hydrogen gas

d. Gaseous ethane (dibarbon hexahydride) reacts with oxygen gas to form gaseous carbon dioxide and gaseous water

e. Solid calcium phosphite reacts with aqueous copper(II) nitrate to form aqueous calcium nitrate and solid copper (II) phosphite

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# 21 CHEMISTRY WORKSHEETS

# WORKED ANSWER KEY

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## **Table of Contents When Is Chemistry Exam**

1. Understanding the eBook When Is Chemistry Exam
  - The Rise of Digital Reading When Is Chemistry Exam
  - Advantages of eBooks Over Traditional Books
2. Identifying When Is Chemistry Exam
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an When Is Chemistry Exam
  - User-Friendly Interface
4. Exploring eBook Recommendations from When Is Chemistry Exam
  - Personalized Recommendations
  - When Is Chemistry Exam User Reviews and Ratings
  - When Is Chemistry Exam and Bestseller Lists
5. Accessing When Is Chemistry Exam Free and Paid eBooks

- When Is Chemistry Exam Public Domain eBooks
- When Is Chemistry Exam eBook Subscription Services
- When Is Chemistry Exam Budget-Friendly Options
- 6. Navigating When Is Chemistry Exam eBook Formats
  - ePub, PDF, MOBI, and More
  - When Is Chemistry Exam Compatibility with Devices
  - When Is Chemistry Exam Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of When Is Chemistry Exam
  - Highlighting and Note-Taking When Is Chemistry Exam
  - Interactive Elements When Is Chemistry Exam
- 8. Staying Engaged with When Is Chemistry Exam
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers When Is Chemistry Exam
- 9. Balancing eBooks and Physical Books When Is Chemistry Exam
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection When Is Chemistry Exam
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine When Is Chemistry Exam
  - Setting Reading Goals When Is Chemistry Exam
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of When Is Chemistry Exam
  - Fact-Checking eBook Content of When Is Chemistry Exam
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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