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ATOMIC NUMBERS, PROTONS, NEUTRONS AND ELECTRONS

The number of protons in each atom of an element is its atomic number. The atomic number is also the number of electrons since atoms themselves are electronically neutral. Because neutrons and protons are almost the same mass, the total number of protons and neutrons in an atom is the atomic mass. Therefore, by subtracting the atomic number from the atomic mass you can calculate the number of neutrons.

This is why the mass number and atomic number, but not the number of neutrons, is shown in the periodic table. Using the information provided below, you can calculate the number of neutrons from the information given.



Example: Helium atom



Use this worksheet and the information provided below, to determine the missing values in the chart below.

Element	Symbol	Atomic Number	Atomic Mass	Protons	Neutrons	Electrons
Hydrogen	H	1	1	1	0	1
Helium	He	2	4	2	2	2
Lithium	Li	3	7	3	4	3
Boron	B	5	11	5	6	5
Carbon	C	6	12	6	6	6
Nitrogen	N	7	14	7	7	7
Oxygen	O	8	16	8	8	8
Fluorine	F	9	19	9	10	9
Neon	Ne	10	20	10	10	10
Sodium	Na	11	23	11	12	11
Magnesium	Mg	12	24	12	12	12
Aluminum	Al	13	27	13	14	13
Silicon	Si	14	28	14	14	14
Phosphorus	P	15	31	15	16	15
Sulfur	S	16	32	16	16	16
Chlorine	Cl	17	35.5	17	18	17
Argon	Ar	18	39.9	18	22	18

Where Are The Electrons Answer Key

Roman Wölfel



Where Are The Electrons Answer Key:

Where Are The Electrons Answer Key Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Where Are The Electrons Answer Key**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve to the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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