

| Question Number | Acceptable Answers  | Reject  | Mark |
|-----------------|---|---------|------|
| 14(a)           | <p>So that the phenol is used up / methyl orange is bleached before the rate changes (significantly)</p> <p>OR</p> <p>So that the phenol is used up / methyl orange is bleached during the initial rate period</p> <p>OR</p> <p>So that the concentration of bromide/bromate/reactants does not fall significantly before all the phenol is used up / the methyl orange is bleached</p> <p>OR</p> <p>Within this region/period/time the average rate of reaction approximates to the initial rate</p> | bromine | (1)  |

| Question Number | Acceptable Answers   | Reject | Mark |
|-----------------|--|--------|------|
| 14(b)(i)        | <p>So that only the <b>concentration</b> of bromide ions varies (significantly) during the course of the reaction / so that the <b>concentration</b> of the bromide ions is the limiting factor / so that the <b>concentration</b> of bromide ions is the only variable</p> <p>ALLOW</p> <p>So their <b>concentrations</b> / the <math>\text{BrO}_3^-</math> and <math>\text{H}^+</math> <b>concentrations</b> do not change</p> <p>OR</p> <p>So their <b>concentrations</b> / the <math>\text{BrO}_3^-</math> and <math>\text{H}^+</math> <b>concentrations</b> are not the limiting factor</p> |        | (1)  |

# Pixl English Past Papers

**M Walker**



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