

1 FOCUS

Objectives

- 1.3.1 **Describe** how Lavoisier transformed chemistry.
- 1.3.2 **Identify** three steps in the scientific method.
- 1.3.3 **Explain** why collaboration and communication are important in science.

Guide for Reading

Build Vocabulary

Word Forms When discussing the two types of variables, explain that one meaning of *manipulate* is “to manage or control” and one meaning of *response* is “to answer or act in turn.”

Reading Strategy

Use Prior Knowledge Before students read the section, ask them what they know about the scientific method.

2 INSTRUCT

Connecting to Your World

Have students read the opening paragraph and look at the photo. Ask, **What made Fleming different from other scientists who had seen this mold?** (Fleming recognized the importance of this discovery. He assumed that the mold had released a chemical that prevented the growth of bacteria.)

Alchemy

Relate

Students who have read the Harry Potter books will be familiar with the name of one alchemist, Nicolas Flamel (1330–1414), who wrote a book about the philosopher’s stone. Alchemists believed that the philosopher’s stone could change base metals into gold.

Guide for Reading



Key Concepts

- How did alchemy lay the groundwork for chemistry?
- How did Lavoisier help to transform chemistry?
- What are the steps in the scientific method?
- What role do collaboration and communication play in science?

Vocabulary

scientific method
observation
hypothesis
experiment
manipulated variable
responding variable
theory
scientific law

Reading Strategy

Building Vocabulary After you read this section, explain the difference between a theory and a scientific law.



20 Chapter 1

Connecting to Your World

In 1928, Alexander Fleming, a Scottish scientist, noticed that a bacteria he was studying did not grow in the presence of a yellow-green mold. Other scientists had made the same observation, but Fleming was the first to recognize its importance. He assumed that the mold had released a chemical that prevented the growth of the bacteria. That chemical was penicillin, which can kill a wide range of harmful bacteria. In 1945, Fleming shared a Nobel Prize for Medicine with Howard Florey and Ernst Chain, who led the team that isolated penicillin. In this section you will study the methods scientists use to solve problems.



Alchemy

The word *chemistry* comes from alchemy. Long before there were chemists, alchemists were studying matter. Alchemy arose independently in many regions of the world. It was practiced in China and India as early as 400 B.C. In the eighth century, Arabs brought alchemy to Spain, from where it spread quickly to other parts of Europe.

Alchemy had a practical side and a mystical side. Practical alchemy focused on developing techniques for working with metals, glass, and dyes. Mystical alchemy focused on concepts like perfection. Because gold was seen as the perfect metal, alchemists were searching for a way to change other metals, such as lead, into gold. Although alchemists did not succeed in this quest, the work they did spurred the development of chemistry.

Alchemists developed the tools and techniques for working with chemicals. Alchemists developed processes for separating mixtures and purifying chemicals. They designed equipment that is still used today, including beakers, flasks, tongs, funnels, and the mortar and pestle in Figure 1.15. What they did not do was provide a logical set of explanations for the changes in matter that they observed. That task was left for chemists to accomplish.

Figure 1.15 A bowl-shaped mortar and a club-shaped pestle are used to grind or crush materials such as herbs, spices, and paint pigments. The mortar and pestle in the photograph is made of porcelain, which is a hard material.

Section Resources

Print

- [Guided Reading and Study Workbook, Section 1.3](#)
- [Core Teaching Resources, Section 1.3 Review](#)
- [Transparencies, T5–T6](#)
- [Laboratory Manual, Lab 1](#)
- [Small-Group Chemistry Laboratory Manual, Lab 1](#)

Technology

- [Interactive Textbook with ChemASAP, Assessment 1.3](#)
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Prentice Hall Chemistry Chapter 1 answers

Albert A Gayle



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Prentice Hall Chemistry, 2000 The Handy Chemistry Answer Book Justin P. Lomont, Ian C. Stewart, 2013-10-01 Don't be mixed up about chemistry Simplify the complex chemical reactions that take place everywhere in our lives with this engaging easy to follow question and answer guide Where would we be without atoms and compounds Gas liquids solids and plasma Acids and bases Bonds and reactions Matter and energy The Handy Chemistry Answer Book covers the building blocks of life and the universe The secret life of atoms how polar bears aren't actually white why oil and water don't mix and much much more are revealed and explained This informative guide covers the basics of chemistry history atomic structures chemical bonds and reactions organic and inorganic chemistry to more advanced material nuclear chemistry biochemistry physical and theoretical chemistry by answering nearly 1 000 common chemistry questions including What causes lightning How does photosynthesis work What are hard and soft Lewis acids and bases What makes a fabric waterproof What are the twelve principles of green chemistry When did alchemists finally abandon trying to make gold What is Le Chatelier's principle What do the different octane ratings mean at the gas pump What is genetic engineering Why is calcium important for strong bones What is the 18 electron rule Why does chocolate turn white as it ages Chemical reactions that rule the world their properties structure composition behavior and history are tackled and explained in plain English in The Handy Chemistry Answer Book With many photos illustrations a few formulas molecular diagrams and other graphics this fun fact filled tome is richly illustrated A history of chemistry timeline appendices on Nobel Prize in Chemistry winners a bibliography further reading section glossary of terms a table of physical constants a table of conversion factors and extensive index add to its usefulness *Computational Chemistry* Errol G. Lewars, 2010-11-10 This corrected second edition contains new material which includes solvent effects the treatment of singlet diradicals and the fundamentals of computational chemistry Computational Chemistry Introduction to the Theory and Applications of Molecular and Quantum Mechanics is an invaluable tool for teaching and researchers alike The book provides an overview of the field explains the basic underlying theory at a meaningful level that is not beyond beginners and it gives numerous comparisons of different methods with one another and with experiment The following concepts are illustrated and their possibilities and limitations are given potential energy surfaces simple and extended Hückel methods ab initio AM1 and related semiempirical methods density functional theory DFT Topics are placed in a historical context adding interest to them and removing much of their apparently arbitrary aspect The large number of references to all significant topics mentioned should make this book useful not only to undergraduates but also to graduate students and academic and industrial researchers Biophysical Chemistry Charles R. Cantor, Paul Reinhard Schimmel, 1980 Biochemistry Donald Voet, Judith G. Voet, 2010-11-16 The Gold Standard in Biochemistry text books Biochemistry 4e is a modern classic that has been thoroughly revised Don and Judy Voet explain biochemical concepts while offering a unified presentation of life and its variation through evolution Incorporates both classical and current

research to illustrate the historical source of much of our biochemical knowledge Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science ,2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction **Aromatic Chemistry** John D. Hepworth, David R. Waring, Michael J. Waring, 2002 This book provides an up to date and comprehensive account of aromatic chemistry A series of chapters describes the synthesis and reactions of the major functional derivatives of benzene and the more common polycyclic systems The concept of aromaticity and the mechanism of aromatic substitution are discussed as is the use of metals in the synthesis of aromatic compounds Throughout emphasis is placed on mechanisms Worked problems and questions are provided to aid understanding In addition to providing material required by an undergraduate studying chemistry Aromatic Chemistry is also ideal for industrial chemists seeking to update their knowledge of this important aspect of chemistry Ideal for the needs of undergraduate chemistry students Tutorial Chemistry Texts is a major new series consisting of short single topic or modular texts concentrating on the fundamental areas of chemistry taught in undergraduate science courses Each book provides a concise account of the basic principles underlying a given subject embodying an independent learning philosophy and including worked examples **Prentice Hall Science Explorer** Michael J. Padilla, 2002 Between Technology and Science Martin Meyer, 2005-02 This dissertation addresses emerging developer communities in a new field of science and technology as well as methods to capture exchange processes between them It contributes to the discussion about a new mode of knowledge production and a changing division of labour between public research industry and government by investigating nanotechnology an emerging area between science and technology To explore exchange processes in this field the study applies various methods In particular it uses patent citation analysis The methodological contribution is a new interpretation of this indicator which sees patent citations as information flows that point to reciprocal exchange processes and potential overlaps between science and technology This is in contrast to the received interpretation which suffers from the application of a framework that was developed in the context of scholarly citation and does not fully appreciate that a patent citation is established by the patent examiner a party external to the inventive process Various formats of patent citation analysis describe nanotechnology as a set of instrument driven scientific fields on their way towards science related technologies Even though nanotechnology patents contain more patent citations to the scientific literature than other technical fields the science and technology systems are relatively autonomous What links them in the case of nano science and technology is a common interest in improving techniques of nano scale measurement and manipulation Another finding is that both countries and firms exhibit relatively strong path dependencies While nanotechnology comprises a key set of

technological areas instrumentation electronics and pharmaceuticals chemicals nano scale activities vary considerably from country to country Also knowledge building activities of firms follow a strong technological path dependency As a result social capital seems to be confined to chiefly technological or scientific trajectories Hence social capital appears not to be very useful in explaining how knowledge is accumulated and integrated at the nano scale Given the central role of instrumentation and the mediated nature of exchange between science and technology at the nano scale public policies should be directed towards supporting education and infrastructure in the area rather than more direct transfer mechanisms

Catalog of Copyright Entries. Third Series Library of Congress. Copyright Office,1972 *Molecular Aggregation* Angelo Gavezzotti,2007 This title provides a brief but accurate summary of all the basic ideas theories methods and conspicuous results of structure analysis and molecular modelling of the condensed phases of organic compounds

Gaither's Dictionary of Scientific Quotations Carl C. Gaither,Alma E. Cavazos-Gaither,2012-01-05 This unprecedented collection of 27 000 quotations is the most comprehensive and carefully researched of its kind covering all fields of science and mathematics With this vast compendium you can readily conceptualize and embrace the written images of scientists laymen politicians novelists playwrights and poets about humankind s scientific achievements Approximately 9000 high quality entries have been added to this new edition to provide a rich selection of quotations for the student the educator and the scientist who would like to introduce a presentation with a relevant quotation that provides perspective and historical background on his subject Gaither s Dictionary of Scientific Quotations Second Edition provides the finest reference source of science quotations for all audiences The new edition adds greater depth to the number of quotations in the various thematic arrangements and also provides new thematic categories *Lasers in Chemistry* David L.

Andrews,2012-12-06 During the three years since the publication of the first edition many applications of lasers in Chemistry have moved across the boundary from academic laboratories to routine instrumental analysis laser mass spectrometry for instance New photochemical techniques have been developed for the study of molecules e g ultrafast spectroscopy In keeping with his successfull concept the author has retained a balance between coverage of more widely used laser methods and new developments

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Multimedia Frank M. Dunnivant, Elliot Anders,2019-05-07 Bridges the gaps between regulatory engineering and science
 disciplines in order to comprehensively cover pollutant fate and transport in environmental multimedia This book presents

and integrates all aspects of fate and transport chemistry modeling various forms of assessment and the environmental legal framework It approaches each of these topics initially from a conceptual perspective before explaining the concepts in terms of the math necessary to model the problem so that students of all levels can learn and eventually contribute to the advancement of water quality science The first third of Pollutant Fate and Transport in Environmental Multimedia is dedicated to the relevant aspects of chemistry behind the fate and transport processes It provides relatively simple examples and problems to teach these principles The second third of the book is based on the conceptual derivation and the use of common models to evaluate the importance of model parameters and sensitivity analysis complex equation derivations are given in appendices Computer exercises and available simulators teach and enforce the concepts and logic behind fate and transport modeling The last third of the book is focused on various aspects of assessment toxicology risk benefit cost and life cycle and environmental legislation in the US Europe and China The book closes with a set of laboratory exercises that illustrate chemical and fate and transport concepts covered in the text with example results for most experiments Features more introductory material on past environmental disasters and the continued need to study environmental chemistry and engineering Covers chemical toxicology with various forms of assessment United States European and Chinese regulations and advanced fate and transport modeling and regulatory implications Provides a conceptual and relatively simple mathematical approach to fate and transport modeling yet complex derivations of most equations are given in appendices Integrates the use of numerous software packages pC pH EnviroLab Simulators Water Wastewater and Global Issues and Fate 2016 Contains numerous easy to understand examples and problems along with answers for most end of the chapter problems and simulators for answers to fate and transport questions Includes numerous companion laboratory experiments with EnviroLab Requiring just a basic knowledge of algebra and first year college chemistry to start Pollutant Fate and Transport in Environmental Multimedia is an excellent textbook for upper level undergraduate and graduate faculty and students studying environmental engineering and science Prentice Hall Exploring Life Science Anthea Maton,1997

Chinese Students and the Experience of International Doctoral Study in STEM Yibo Yang, Judith

MacCallum,2022-09-30 This volume examines the diversified and challenging experiences of Chinese international STEM doctoral students at Australian institutes of higher education exploring how intersections between research personal life and social experiences can be negotiated to achieve academic success and personal transformation By drawing on a range of qualitative and longitudinal research methods the book foregrounds student narratives and utilizes a novel three dimensional multi world framework as an effective approach for understanding student experiences in a holistic way It integrates Chinese philosophical perspectives and theories in the fields of educational psychology international education and doctoral education to interpret the nuances complexity and particularities of the cross cultural STEM PhD experience highlighting the importance of the supervisor mentee relationship and the role of students cultural social and philosophical values in

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