

# Electrons in Atoms

## Section 5.1 Light and Quantized Energy

In your textbook, read about the wave nature of light.

Use each of the terms below just once to complete the passage.

amplitude  
light

energy  
wave

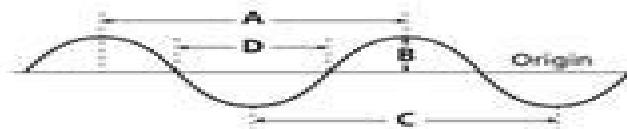
frequency  
wavelength

hertz  
speed

Electromagnetic radiation is a kind of (1) \_\_\_\_\_ that behaves like a(n) (2) \_\_\_\_\_ as it travels through space. (3) \_\_\_\_\_ is one type of electromagnetic radiation. Other examples include X rays, radio waves, and microwaves.

All waves can be characterized by their wavelength, amplitude, frequency, and (4) \_\_\_\_\_. The shortest distance between equivalent points on a continuous wave is called a(n) (5) \_\_\_\_\_. The height of a wave from the origin to a crest or from the origin to a trough is the (6) \_\_\_\_\_. (7) \_\_\_\_\_ is the number of waves that pass a given point in one second. The SI unit for frequency is the (8) \_\_\_\_\_, which is equivalent to one wave per second.

Use the figure to answer the following questions.



9. Which letter(s) represent one wavelength? \_\_\_\_\_
10. Which letter(s) represent the amplitude? \_\_\_\_\_
11. If twice the length of A passes a stationary point every second, what is the frequency of the wave? \_\_\_\_\_

# Prentice Hall Chemistry Chapter Test

**Kelly M. Elkins**

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**CRC Handbook of Basic Tables for Chemical Analysis** Thomas J. Bruno,Paris D.N. Svoronos,2020-07-30 Researchers in chemistry chemical engineering pharmaceutical science forensics and environmental science make routine use of chemical analysis but the information these researchers need is often scattered in different sources and difficult to access The CRC Handbook of Basic Tables for Chemical Analysis Data Driven Methods and Interpretation Fourth Edition is a one stop reference that presents updated data in a handy format specifically designed for use when reaching a decision point in designing an analysis or interpreting results This new edition offers expanded coverage of calibration and uncertainty and continues to include the critical information scientists rely on to perform accurate analysis Enhancements to the Fourth Edition Compiles a huge array of useful and important data into a single convenient source Explanatory text provides context for data and guidelines on applications Coalesces information from several different fields Provides information on the most useful wet chemistry methods as well as instrumental techniques with an expanded discussion of laboratory safety Contains information of historical importance necessary to interpret the literature and understand current methodology Unmatched in its coverage of the range of information scientists need in the lab this resource will be referred to again and again by practitioners who need quick easy access to the data that forms the basis for experimentation and analysis

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led the development of modern forensic analytic techniques and practices for use in a variety of applications Introduction to Forensic Chemistry is the perfect balance of testing methods and application Unlike other competing books on the market coverage is neither too simplistic nor overly advanced making the book ideal for use in both undergraduate and graduate courses The book introduces chemical tests spectroscopy advanced spectroscopy and chromatography to students The second half of the book addresses applications and methods to analyze and interpret controlled substances trace evidence questioned documents firearms explosives environmental contaminants toxins and other topics The book looks at innovations in the field over time including the latest development of new discernible chemical reactions instrumental tools methods and more Key features Nearly 300 full color figures illustrating key concepts and over 20 case studies Addresses all the essential topics without extraneous or overly advanced coverage Includes full pedagogy of chapter objectives key terms lab problems end of chapter questions and additional readings to emphasize key learning points Includes chemical structures and useful spectra as examples Fulfils the forensic chemistry course requirement in FEPAC accredited programs Includes a chapter on Chemical Biological Radiological Nuclear and Explosive CBRNE materials Comprehensive and accessible without being overly technical Introduction to Forensic Chemistry will be a welcome addition to the field and an ideal text designed for both the student user and professor in mind Course ancillaries including an Instructor s Manual with Test Bank and chapter PowerPoint lecture slides are available with qualified course adoption

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### **Handbook of**

***Isolation and Characterization of Impurities in Pharmaceuticals*** Satinder Ahuja,Karen Mills Alsante,2003-07-18 The United States Food and Drug Administration FDA and other regulatory bodies around the world require that impurities in drug

substance and drug product levels recommended by the International Conference on Harmonisation ICH be isolated and

characterized Identifying process related impurities and degradation products also helps us to understand the production of impurities and assists in defining degradation mechanisms When this process is performed at an early stage there is ample time to address various aspects of drug development to prevent or control the production of impurities and degradation products well before the regulatory filing and thus assure production of a high quality drug product This book therefore has

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