

1B-2 Practice Problems: Mass-spectrometry, empirical and molecular formulae

1. Calculate percentage composition of a hydrocarbon mixture A (molecular mass 100 g mol⁻¹) and B (molecular mass 120 g mol⁻¹) from a full spectrum of mass-spectrometry.
2. Calculate the relative molecular mass of a hydrocarbon from its mass-spectrometry data. Calculate the empirical formula from mass-spectrometry data. Calculate the molecular formula from mass-spectrometry data.
3. Calculate percentage composition of a hydrocarbon mixture A (molecular mass 100 g mol⁻¹) and B (molecular mass 120 g mol⁻¹) from a full spectrum of mass-spectrometry.
4. Calculate the relative molecular mass of a hydrocarbon from its mass-spectrometry data. Calculate the empirical formula from mass-spectrometry data. Calculate the molecular formula from mass-spectrometry data.
5. Calculate percentage composition of a hydrocarbon mixture A (molecular mass 100 g mol⁻¹) and B (molecular mass 120 g mol⁻¹) from a full spectrum of mass-spectrometry.
6. Calculate the relative molecular mass of a hydrocarbon from its mass-spectrometry data. Calculate the empirical formula from mass-spectrometry data. Calculate the molecular formula from mass-spectrometry data.

Empirical and Molecular Formulae

1. Calculate the relative molecular mass of a hydrocarbon from its mass-spectrometry data. Calculate the empirical formula from mass-spectrometry data. Calculate the molecular formula from mass-spectrometry data.
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Christian G. Meyer



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