

acetate	$\text{C}_2\text{H}_3\text{O}_2^-$
ammonium	NH_4^+
arsenate	AsO_4^{3-}
arsenite	AsO_3^{3-}
azide	N_3^-
benzoate	$\text{C}_7\text{H}_5\text{O}_2^-$
borate	BO_3^{3-}
bromate	BrO_3^-
carbonate	CO_3^{2-}
chlorate	ClO_3^-
chlorite	ClO_2^-
chromate	CrO_4^{2-}
cyanide	CN^-
dichromate	$\text{Cr}_2\text{O}_7^{2-}$
dihydrogen phosphate	H_2PO_4^-
dihydrogen phosphite	H_2PO_3^-
hydrogen carbonate	HCO_3^-
hydrogen phosphate	HPO_4^{2-}
hydrogen phosphite	HPO_3^{2-}
hydrogen sulfate	HSO_4^-
hydrogen sulfide	HS^-
hydrogen sulfite	HSO_3^-
hydroxide	OH^-
hypochlorite	ClO^-
iodate	IO_3^-
manganate	MnO_4^{2-}
nitrate	NO_3^-
nitrite	NO_2^-
oxalate	$\text{C}_2\text{O}_4^{2-}$
perchlorate	ClO_4^-
permanganate	MnO_4^-
peroxide	O_2^{2-}
phosphate	PO_4^{3-}
phosphite	PO_3^{3-}
silicate	SiO_3^{2-}
sulfate	SO_4^{2-}
sulfite	SO_3^{2-}
tartrate	$\text{C}_4\text{H}_4\text{O}_6^{2-}$
thiocyanate	SCN^-
thiosulfate	$\text{S}_2\text{O}_3^{2-}$

AsO_3^{3-}	arsenite
AsO_4^{3-}	arsenate
BO_3^{3-}	borate
BrO_3^-	bromate
$\text{C}_2\text{H}_3\text{O}_2^-$	acetate
$\text{C}_2\text{O}_4^{2-}$	oxalate
$\text{C}_4\text{H}_4\text{O}_6^{2-}$	tartrate
$\text{C}_7\text{H}_5\text{O}_2^-$	benzoate
ClO^-	hypochlorite
ClO_2^-	chlorite
ClO_3^-	chlorate
ClO_4^-	perchlorate
CN^-	cyanide
CO_3^{2-}	carbonate
$\text{Cr}_2\text{O}_7^{2-}$	dichromate
CrO_4^{2-}	chromate
H_2PO_3^-	dihydrogen phosphite
H_2PO_4^-	dihydrogen phosphate
HCO_3^-	hydrogen carbonate
HPO_3^{2-}	hydrogen phosphite
HPO_4^{2-}	hydrogen phosphate
HS^-	hydrogen sulfide
HSO_3^-	hydrogen sulfite
HSO_4^-	hydrogen sulfate
IO_3^-	iodate
MnO_4^-	permanganate
MnO_4^{2-}	manganate
N_3^-	azide
NH_4^+	ammonium
NO_2^-	nitrite
NO_3^-	nitrate
O_2^{2-}	peroxide
OH^-	hydroxide
PO_3^{3-}	phosphite
PO_4^{3-}	phosphate
$\text{S}_2\text{O}_3^{2-}$	thiosulfate
SCN^-	thiocyanate
SiO_3^{2-}	silicate
SO_3^{2-}	sulfite
SO_4^{2-}	sulfate

Poly Atomic Ions Packet

G Psacharopoulos



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Super-Intense Laser-Atom Physics IV H.G. Muller, M.V. Fedorov, 1996-05-31 Atoms in strong radiation fields are interesting objects for study and the research field that concerns itself with this study is a comparatively young one For a long period after the discovery of the photoelectric effect it was not possible to generate electro magnetic fields that did more than perturb the atom only slightly and first order perturbation theory could perfectly explain what was going on at those low intensities The development of the pulsed laser has changed this state of affairs in a rather dramatic way and fields can be applied that really have a large or even dominant influence on atomic structure In the latter case we speak of super intense fields Since the interaction between atoms and electromagnetic waves is characterized by many parameters other than the light intensity such as frequency ionization potential orbit time etc it is actually quite difficult to define what is exactly meant by the term super intense Obviously the term does not have an absolute meaning and intensity should always be viewed in relation to other properties of the system An atom in a radiation field can thus best be described in terms of various ratios of the quantities involved The nature of the system sometimes drastically changes if the value of one of these parameters exceeds a certain critical value and the new regime could be called super intense with respect to that parameter

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including limitations current problems and areas in which the method does not succeed well **Physical Science** C. Lon Enloe, Elizabeth Garnett, Jonathan Miles, Stephen Swanson, 2000-12-25 This is an introductory book that provides students with the tools to master the basic principles of physics and chemistry needed by the aspiring technology professional Like all the books in the critically acclaimed Preserving the Legacy series each chapter is divided into subsections featuring learning objectives and a Check Your Understanding section to help students focus on important concepts Questions requiring written and mathematical answers at the end of each chapter provide students with the opportunity to further demonstrate their understanding of the concepts The only book available that specifically addresses the emerging need for a course to teach physics and chemistry principles to the growing number of students entering the various fields of technology it offers a thorough grounding in foundational concepts along with Technology boxes that offer practical applications Physical Science What the Technology Professional Needs to Know features Crucial topics such as measuring systems matter energy motion electricity and magnetism electromagnetic radiation nuclear radiation and reactions and chemical reactions and solutions Integrated coverage linking specific concepts to everyday applications An extensive glossary offering quick access to essential terminology An accompanying laboratory manual with additional exercises to enhance learning With its comprehensive coverage and quick reference format Physical Science What the Technology Professional Needs to Know is also a handy resource for any technology professional needing a quick refresher or useful working reference *Physics & Chemistry of Fullerenes* Peter W. Stephens, 1993 The creation of the hollow carbon buckminsterfullerene molecule as well as methods to produce and purify bulk quantities of it has triggered an explosive growth of research in the field Superconducting and magnetic fullerenes atoms trapped inside the fullerene cage chemically bonded fullerene complexes and nanometer scale helical carbon tubes are some of the leading areas that have generated much excitement This book is intended as a guide to the literature for the scientist who is just entering fullerene research and will be one more valuable volume to the collection for the established worker It contains reprints of some sixty most important research papers with focus especially on those papers that have guided further work in the field There is also a short review of the field with references to many other publications CliffsNotes AP Chemistry Angela Woodward Spangenberg, 2016-01-12 Test prep for the AP Chemistry exam with 100% brand new content that reflects recent exam changes Addressing the major overhaul that the College Board recently made to the AP Chemistry exam this AP Chemistry test prep guide includes completely brand new content tailored to the exam administered every May Features of the guide include review sections of the six big ideas that the new exam focuses on Fundamental building blocks Molecules and interactions Chemical reactions Reaction rates Thermodynamics Chemical equilibrium Every section includes review questions and answers Also included in the guide are two full length practice tests as well as a math review section and sixteen discrete laboratory exercises to prepare AP Chemistry students for the required laboratory experiments section on the exam Comprehensive Analytical Chemistry

Cecil Leeburn Wilson, David Woodburn Wilson, 2003 *Physics Briefs* , 1994 *The Journal of Chemical Physics* , 2008

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