

- a) What are Carbenes ? What is the state of their hybridisation ? Discuss structure and stability of singlet and triplet dichlorocarbene. 5
- b) Briefly explain the following observations : 2×2
- Phenyl anion is more stable than cyclohexyl anion.
 - N, N-dimethylaniline is weak as a base than N, N, Z, 6 - tetramethylaniline.
- c) Methane reacts with Cl_2 to form CH_3Cl and HCl , but with I_2 doesn't occur at appreciable rate, even $\text{I} - \text{I}$ bond is much weaker than $\text{Cl} - \text{Cl}$ bond. Explain why this is SO. 3
- d) What products are formed when glycine ($\text{H}_2\text{NCH}_2\text{CO}_2\text{H}$) is treated with conc. HCl and NaOH separately. 3
2. a) Draw the structure for each of the following compounds : $1\frac{1}{2} \times 4$
- (1R, 3S) - 1, 3-cyclohexanediol
 - (1Z, 3E) - 1-bromo-1, 3-pentadiene.

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