

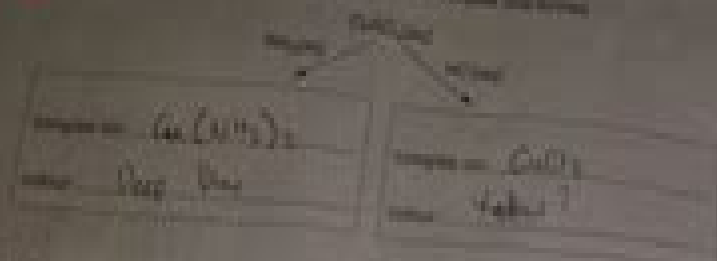
Question 1

Answer all the questions

The reaction is about the synthesis of copper complexes and ligands.

(a) The reaction shows the synthesis of a copper complex.

In the boxes, write the formula and name of the ligands and metal.



(b) Cu^{2+} has two d-orbitals in d_{sp^2} hybridization state and two water molecules. The d_{sp^2} hybridization is a tetrahedral ligand.

The resulting formula of the complex is shown in Fig. 1.1 below.



Fig. 1.1

(c) What is meant by the term tetrahedral ligand?

A ligand that forms four bonds with central metal ion.

(d) The complex is a copper(II) complex.

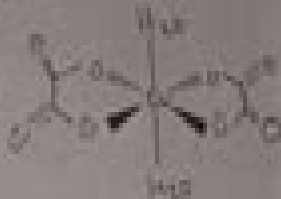

The charge is +2 of the complex is a copper(II) complex. The charge has been given.



Complex A

Complete the 3D diagram of the other two geometries of A. You do not need to include any charges.

Indicate with ticks whether the stereoisomers are cis, trans, optical or a combination of these types.

Stereoisomer													
	<table><tr><td>cis</td><td><input type="checkbox"/></td></tr><tr><td>trans</td><td><input checked="" type="checkbox"/></td></tr><tr><td>optical</td><td><input type="checkbox"/></td></tr></table>	cis	<input type="checkbox"/>	trans	<input checked="" type="checkbox"/>	optical	<input type="checkbox"/>	<table><tr><td>cis</td><td><input checked="" type="checkbox"/></td></tr><tr><td>trans</td><td><input type="checkbox"/></td></tr><tr><td>optical</td><td><input checked="" type="checkbox"/></td></tr></table>	cis	<input checked="" type="checkbox"/>	trans	<input type="checkbox"/>	optical
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(e) What is the empirical formula, including the charge, of the complex in A?

$[Cu(NH_3)_2(H_2O)_2]^{2+}$

Ocr Chemistry A2 Unofficial Mark Scheme 2014

Siddappa N.Byrareddy



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 - The Rise of Digital Reading Ocr Chemistry A2 Unofficial Mark Scheme 2014
 - Advantages of eBooks Over Traditional Books
2. Identifying Ocr Chemistry A2 Unofficial Mark Scheme 2014
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 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ocr Chemistry A2 Unofficial Mark Scheme 2014
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ocr Chemistry A2 Unofficial Mark Scheme 2014
 - Personalized Recommendations
 - Ocr Chemistry A2 Unofficial Mark Scheme 2014 User Reviews and Ratings
 - Ocr Chemistry A2 Unofficial Mark Scheme 2014 and Bestseller Lists

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 - Ocr Chemistry A2 Unofficial Mark Scheme 2014 Budget-Friendly Options
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 - ePub, PDF, MOBI, and More
 - Ocr Chemistry A2 Unofficial Mark Scheme 2014 Compatibility with Devices
 - Ocr Chemistry A2 Unofficial Mark Scheme 2014 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ocr Chemistry A2 Unofficial Mark Scheme 2014
 - Highlighting and Note-Taking Ocr Chemistry A2 Unofficial Mark Scheme 2014
 - Interactive Elements Ocr Chemistry A2 Unofficial Mark Scheme 2014
8. Staying Engaged with Ocr Chemistry A2 Unofficial Mark Scheme 2014
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ocr Chemistry A2 Unofficial Mark Scheme 2014
9. Balancing eBooks and Physical Books Ocr Chemistry A2 Unofficial Mark Scheme 2014
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ocr Chemistry A2 Unofficial Mark Scheme 2014
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Ocr Chemistry A2 Unofficial Mark Scheme 2014
 - Setting Reading Goals Ocr Chemistry A2 Unofficial Mark Scheme 2014
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Ocr Chemistry A2 Unofficial Mark Scheme 2014
 - Fact-Checking eBook Content of Ocr Chemistry A2 Unofficial Mark Scheme 2014
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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