

EXPERIMENT 7: REDOX TITRATION

PURPOSE

The purpose of this experiment is to standardize the potassium permanganate solution by titration method.

INTRODUCTION

Oxidation-reduction processes electrons transfer one species to another. Potassium permanganate has been widely used as an oxidizing agent and requires no indicator unless very dilute solutions are used. The oxidizing action of KMnO_4 in the acidic medium can be represented by the following equation:



Solution containing MnO_4^- ions are purple in colour and the solution containing Mn^{2+} ions colourless and hence permanganate solution is decolourised when added to a solution of a reducing agent.

CHEMICALS AND APPARATUS

Potassium permanganate solution	Burette
Dried sodium oxalate	Conical flask
1M sulphuric acid	Hot plate
	Pipette
	Thermometer

PROCEDURE

1. The burette was rinsed and filled with potassium permanganate solution. the air bubble was removed from the nozzle of the burette by draining some of the solutions..
2. 0.3 g to 0.4 g of dried sodium oxalate was weighed accurately. The mass of sodium oxalate was noted and transferred to conical flask.
3. 40 ml of 1.0 M sulphuric acid was added to the conical flask and carefully heated to 60°C .

Redox Titration Experiment Report Guidelines

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The Analytical Chemistry Laboratory Companion Michael D. Holloway, 2025-07-15 The Analytical Chemistry Laboratory Companion is essential for both students and professionals as it provides quick clear explanations on critical topics in analytical chemistry equipping you with the statistical tools necessary to ensure accurate and reliable data interpretation The Analytical Chemistry Laboratory Companion serves as a reference guide for students and professionals alike who need quick explanations on specific topics laboratory operations the structure of designing experiments and the use of statistics to gain increased accuracy precision repeatability and reproducibility of data This volume will also provide in depth and advanced studies and build the necessary background knowledge for success in the field This companion provides a concise examination of the various analytical tools used for chemistry and defines basic analytical instrument principles techniques and applications in addition to exploring statistical tools useful in data interpretation test result reporting and common root causes for faulty data with suggested remedies The introduction provides a concise guide on foundational topics such as developing standard operating procedures laboratory safety instrumental analytical methods and common statistical tools useful for data interpretation This companion covers both wet chemical and instrumental analysis including their principles applications and pitfalls The Analytical Chemistry Laboratory Companion is a must have comprehensive guide in the field of analytical chemistry

Illustrated Guide to Home Chemistry Experiments Robert Bruce Thompson, 2012-02-17 For students DIY hobbyists and science buffs who can no longer get real chemistry sets this one of a kind guide explains how to set up and use a home chemistry lab with step by step instructions for conducting experiments in basic chemistry not just to make pretty colors and stinky smells but to learn how to do real lab work Purify alcohol by distillation Produce hydrogen and oxygen gas by electrolysis Smelt metallic copper from copper ore you make yourself Analyze the makeup of seawater bone and other common substances Synthesize oil of wintergreen from aspirin and rayon fiber from paper Perform forensics tests for fingerprints blood drugs and poisons and much more From the 1930s through the 1970s chemistry sets were among the most popular Christmas gifts selling in the millions But two decades ago real chemistry sets began to disappear as manufacturers and retailers became concerned about liability

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