# 2002 Edition • Reference Tables for Physical Setting/Chemistry

#### Table A Standard Temperature and Pressure

Name	Value	Umit
Standard Pressure	101,3 kPa 1 atm	kilopascal atmosphere
Standard Temperature	273 K 0°C	kelvin degree Celsius

# Table B Physical Constants for Water

Heat of Fusion	333.6 J/g
Heat of Vaporization	2259 J/g
Specific Heat Capacity of H <sub>2</sub> O (ℓ)	4.2 J/g•K

#### Table D Selected Units

Symbol	Name	Quantity
m	meter	length
kg	kilogram	THASS
. Pa.	pascal	pressure
K	kelvin	temperature
mol	mole	amount of substance
J	joule	energy, work, quantity of beat
260	second	time
L	liter	volume
ppan	part per million	concentration
M	molarity	solution concentration

#### Table C Selected Prefixes

Factor.	Prefix	Symbol
$10^{3}$	kilo-	k
$10^{-1}$	deci-	d
10-2	centi-	C
10-3	málli-	m
10-6	micro-	54
10-9	nano-	n
$10^{-12}$	pico-	P

Table E Selected Polyatomic Ions

H <sub>3</sub> O <sup>+</sup>	hydronium	CrO <sub>4</sub> 2-	chromate
$Hg_2^{-2+}$	dimercury (I)	$Cr_2O_7^{2-}$	dichromate
NH <sub>4</sub> <sup>+</sup>	ammonium	$MmO_4^-$	permanganate
$C_2H_3O_2^-$ $CH_3COO^-$ acetate	} acetate	$NO_2^-$	nitrite
	) · J	NO <sub>3</sub> -	nitrate
CN-	cyanide	O22-	peroxide
CO <sub>3</sub> <sup>2</sup> -	carbonate	OH-	hydroxide
HCO <sub>3</sub>	hydrogen carbonate	PO <sub>4</sub> 3-	phosphate
$C_2O_4^{2-}$	oxalate	SCN-	thiocyanate
clo-	hypochlorite	$SO_3^{2-}$	sulfite
ClO <sub>2</sub>	chlorite	SO <sub>4</sub> 2-	sulfate
ClO <sub>3</sub> <sup>-</sup>	chlorate	HSO <sub>4</sub>	hydrogen sulfate
ClO <sub>4</sub> =	perchlorate	S <sub>2</sub> O <sub>3</sub> 2-	thiosulfate

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