



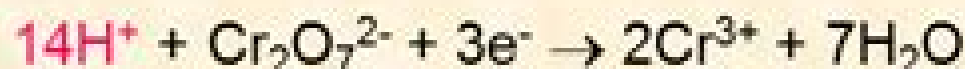
Step 2: Balance each kind of atom other than H and O



Step 3: Balance oxygen atoms by using  $\text{H}_2\text{O}$



Step 4: Balance H atoms by using  $\text{H}^+$  ions



Step 5: Use electrons as needed to obtain a charge that is balanced



+2

+2

+9

+6

Add three electrons to the reactant side to balance the charges



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