

## 5-2 Review and Reinforcement

### Reading the Periodic Table

On the line at the left, write the letter of the approximate location of each group of elements on the periodic table below. Some letters will be used more than once.

- |                                  |   |
|----------------------------------|---|
| _____ 1. carbon family           | _____ 8. f-block elements                       |
| _____ 2. alkaline earth metals   | _____ 9. noble gases                            |
| _____ 3. inner transition metals | _____ 10. p-block elements                      |
| _____ 4. halogens                | _____ 11. nitrogen family                       |
| _____ 5. d-block elements        | _____ 12. s-block elements                      |
| _____ 6. oxygen group            | _____ 13. transition metals                     |
| _____ 7. alkali metals           | _____ 14. group of one nonmetal and four metals |



Use the skills you developed in Section 5-2 to answer each of the following questions. Below is the abbreviated electron configuration for sodium. Explain each part of this configuration in the space provided.

- $[Ne] 3s^1$
- |           |
|-----------|
| 15. _____ |
| 16. _____ |
| 17. _____ |
| 18. _____ |

# Reinforcement 6 Answer Key

**Gregory J. Madden, Derek D.  
Reed, Florence D. DiGennaro Reed**

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