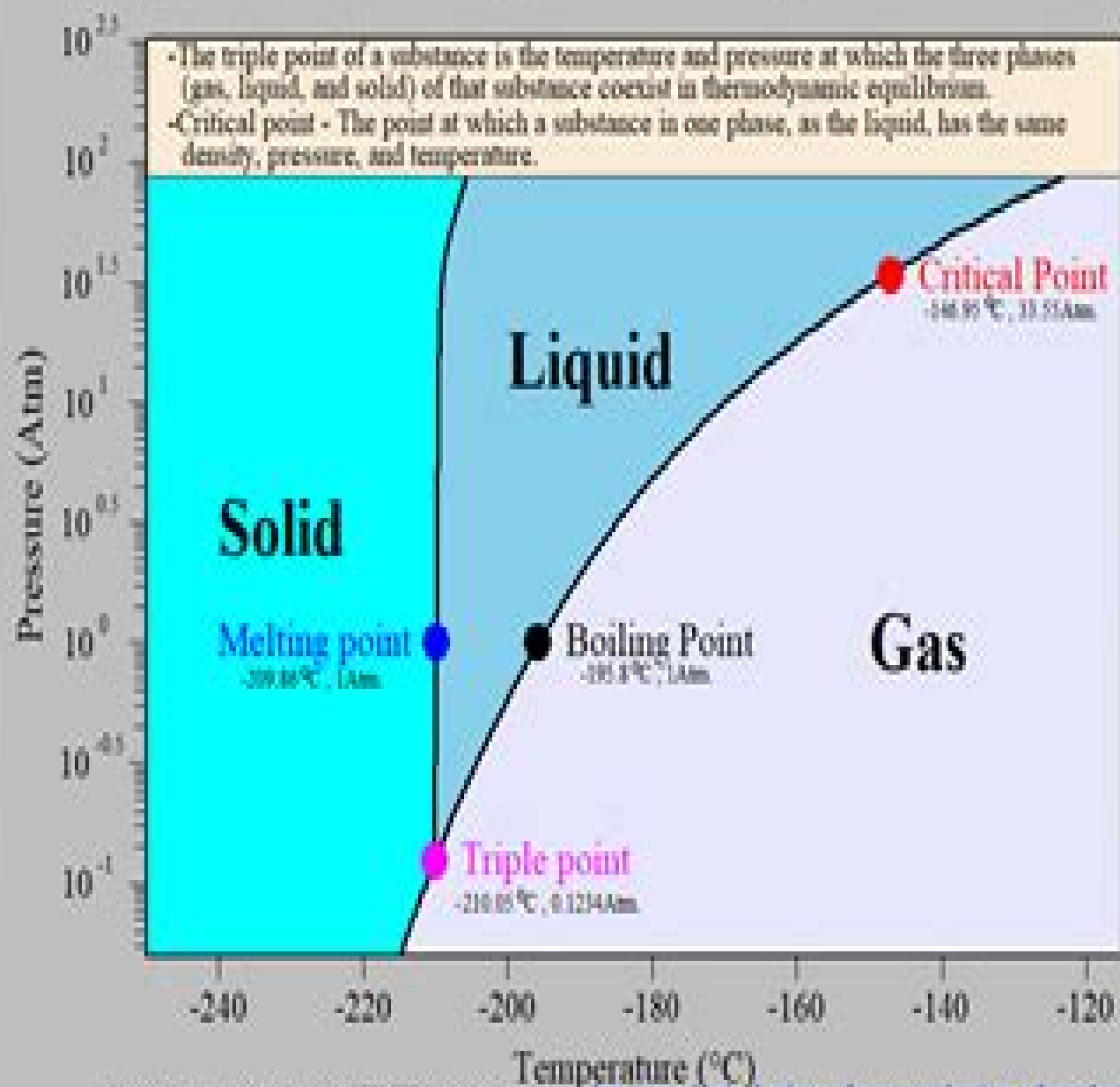


Nitrogen Phase Diagram

- The triple point of a substance is the temperature and pressure at which the three phases (gas, liquid, and solid) of that substance coexist in thermodynamic equilibrium.
- Critical point - The point at which a substance in one phase, as the liquid, has the same density, pressure, and temperature.



Title Nitrogen Phase Diagram

X and Y axis Temperature (°C) Pressure (Atm)

X min	X max	Y min	Y max
-250	-115	0.05	100

Generate plot **File**

Interpolate **Drawn width** **Black** **Width**

X Y Text size **Symbol size**

170 0.05 [-0]
-115 100 [-0]

1072.7962 - The triple point of a substance is the temperature and pressure at which the three phases coexist in thermodynamic equilibrium [-0]
1072.7962 (gas, liquid, and solid) of that substance coexist in thermodynamic equilibrium [-0]
1071.4582 - Critical point - The point at which a substance in one phase, as the liquid, has the same [-0]

X log() **Y log()** **Calculate** **Data set**

Tick marks **Legend size** **Symbol** **Fill color** **Frame**

Legend text

Options/Polymath **Name - Experiment**

nitro/polymath **Nitrogen Phase Diagram - First (any mode)**

Width **Nitrogen Phase Diagram - First (any mode)**

Connect **Polymath** **Nitrogen Phase Diagram - First (any mode)**

Delete **Equilibrium phase diagram of nitrogen**

1 nitro/polymath **Solid - Symbol phase map - Phase diagram**

Black **Solid - Phase diagram interpolated**

Grey mode **Double - Phase diagram interpolated**

Options **Volume - Phase diagram**

Fill colors **Carbon dioxide - water phase diagram**

Options/Polymath **P-H diagram - Volume**

Interpolate points **Edit list box**

View coordinates **Save**

Stop color fill by interpolation lines **Before** **After**

Grid **Off** **On** **Replace** **Delete**

Scale **Off** **On**

Frame **Off** **On**

Phase Diagram Oxygen Nitrogen

Sabine Zange

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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