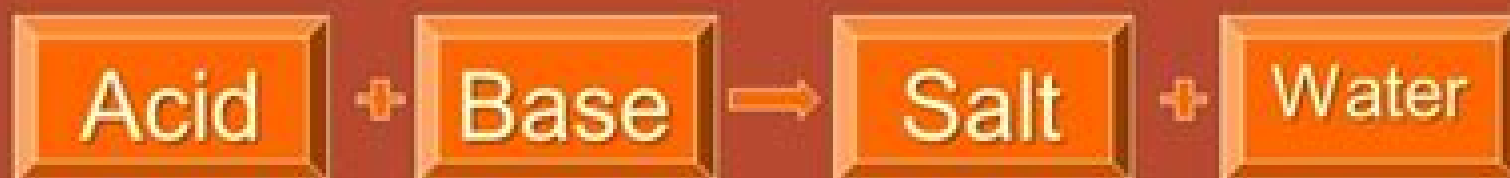


# Neutralisation is exothermic



*Acid + base reactions are called neutralisation reactions because the low pH (<7) of an acid and the high pH (>7) of a base, results in a pH closer to neutral (7)*

*These reactions release heat and are exothermic. For example:*



# Neutralisation Exothermic Results

**CH Cherryholmes**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the 'CH Cherryholmes' text.

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