

Write the electron configuration for the following elements:

- a) Magnesium _____
- b) Cobalt _____
- c) Krypton _____
- d) Beryllium _____
- e) Scandium _____
- f) Nickel _____
- g) Cadmium _____
- h) Selenium _____
- i) Strontium _____
- j) Lithium _____

Determine what elements are denoted by the following electron configurations:

- a) $1s^2 2s^2 2p^4 3s^2 3p^2$ _____
- b) $1s^2 2s^2 2p^6 3s^2 3p^4 4s^2 3d^{10} 4p^4 5s^2$ _____
- c) $1s^2 2s^2 2p^2$ _____
- d) $1s^2 2s^2 2p^6 3s^2 3p^4 4s^2$ _____
- e) $1s^2 2s^2 2p^6 3s^2 3p^4 4s^2 3d^6$ _____
- f) $1s^2 2s^2 2p^6 3s^2 3p^4 4s^2 3d^{10} 4p^4 5s^2 4d^{10} 5p^3$ _____
- g) $1s^2 2s^2 2p^6 3s^2$ _____
- h) $1s^1$ _____
- i) $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^2$ _____
- j) $1s^2 2s^2$ _____

Determine whether the following electron configuration are or are not valid:

- a) $1s^2 2s^2 2p^6 3s^2 4s^2 4d^{10} 4p^4$ _____
- b) $1s^2 2s^2 2p^6 3s^2 3p^4$ _____
- c) $1s^2 2s^2 3s^2 2p^6 4s^2$ _____
- d) $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2$ _____

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